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# **Investigation of Structure and Dynamics in Polymeric Systems via Solid State NMR**

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Robert Graf

*Max-Planck-Institut für Polymerforschung  
Mainz*

March 2<sup>nd</sup>, 2005





founded 1983, 450 co-workers, on campus of the University of Mainz,



interdisciplinary fundamental research of polymers

# Board of Directors

Materials  
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Polymer  
Analysis

Surfaces  
and Interfaces

## Scientific Project Areas

Development  
of Methods

X-Ray Structure  
Analysis

Structure  
and Dynamics

Functional  
Materials

Computer and  
Network Service

Mechanical and  
Dielectric  
Relaxation

Clean Room



# Max Planck Institute for Polymer Research

# Scientific Activities of the MPI-P Spectroscopy Group



## Structure and Dynamics

- Polymer Chain Dynamics
- Local-Order Phenomena

## Polymer Synthesis

- Polymerization:  
Radical, Anionic, Emulsion,  
Atom-Transfer Radical
- Isotopic Labelling

## Functional Materials

- Photoconductors
- Protonconductors
- Shape-Persistent Macromolecules

## Development of Methods

- 2D Multiple Quantum MAS NMR
- Imaging of Hyperpolarized Gases
- DFT Calculations
- Pulsed and CW EPR
- Fourier-Transform Rheology

## Spectroscopy Group

- Polyelectrolyte Multilayers
- Self Assembled Monolayers
- Surface Patterning

## Surfaces & Interfaces

- Columnar  $\pi-\pi$  Stacking
- Hydrogen-Bonded Structures
- Coordination Polymers

## Supramolecular Architectures





# Investigation of Structure and Dynamics in Polymeric Systems via Solid State NMR

**Introduction** • interactions in solid state NMR

**Solid State NMR** • resolution enhancement in solid state NMR,  
magic angle spinning, recoupling methods,  
double quantum NMR spectroscopy

**Polymer dynamics** • Polyelectrolyte layers, polybutadiene, PEMA

**Conclusions** • Pro and Contra of Solid State NMR investigations



# Molecular Structures and Dynamics via NMR



Important NMR interactions:

$$\mathbf{H} = \mathbf{H}_Z + \mathbf{H}_Q + \mathbf{H}_{CS} + \mathbf{H}_D + \mathbf{H}_J$$

Zeemann Interaction :

$$\mathbf{H}_Z = -\sum_i \gamma_i \underline{B}_0 \underline{\mathbf{I}}^i$$

Quadrupol Interaction :

$$\mathbf{H}_Q = -\sum_i \frac{eQ}{2I(2I-1)\hbar} \underline{\mathbf{I}}^i \underline{\underline{\mathbf{V}}} \underline{\mathbf{I}}^i$$

Electronic Shielding :

$$\mathbf{H}_{CS} = -\sum_i \gamma_i \underline{B}_0 \underline{\underline{\boldsymbol{\sigma}}} \underline{\mathbf{I}}^i$$

Dipol-Dipol Interaction :

$$\mathbf{H}_D = -\sum_{i \neq j} \frac{\mu_0 \hbar}{4\pi} \frac{\gamma_i \gamma_j}{r^3} \left[ \frac{3}{r^2} (\underline{\mathbf{I}}^i \cdot \underline{\mathbf{r}}) (\underline{\mathbf{I}}^j \cdot \underline{\mathbf{r}}) - \underline{\mathbf{I}}^i \cdot \underline{\mathbf{I}}^j \right]$$

Indirect Spin-Spin Interaction :  $\mathbf{H}_J = -\sum_{i \neq j} \underline{\mathbf{I}}^i \cdot \underline{\underline{\mathbf{J}}}^{ij} \underline{\mathbf{I}}^j$



# Interactions in Solid State NMR Spectroscopy



Zeemann interaction dominates all other NMR interactions

Perturbation  
Theory

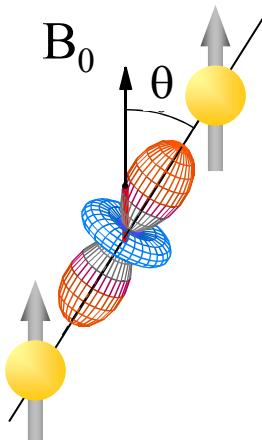
Orientation dependence of local spin interaction on  $B_0$

## Isotropic Contributions

$H_{cs}$ : chemical shift

$H_J$ : J-couplings

## Liquid state NMR



$H_Q, H_D, H_{cs}, H_J$

## Anisotropic Contributions

### Symmetric

$H_Q$ : quadrupol

$H_D$ : dipol-dipol

$H_{cs}$ : chemical shift

### Asymmetric

$H_Q$ : quadrupol

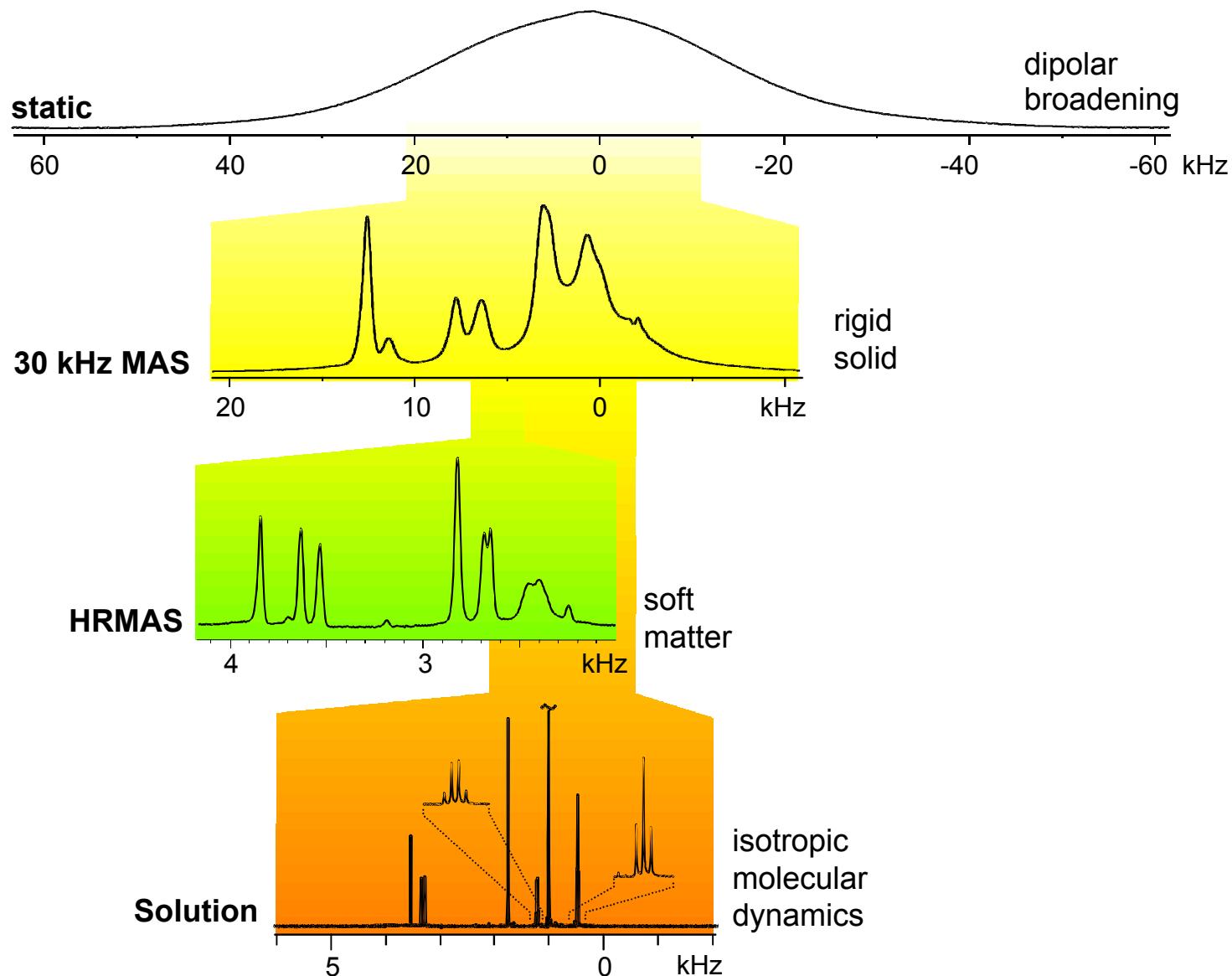
$H_{cs}$ : chemical shift

$$H_D = \sum_{i \neq j} \frac{\mu_0 \hbar}{4\pi} \frac{\gamma_i \gamma_j}{r_{ij}^3} \frac{1}{2} (3 \cos^2 \theta_{ij} - 1) T_{2,0}^{ij}$$

*Distance      Orientation*



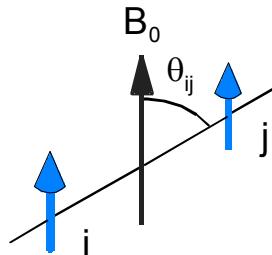
# $^1\text{H}$ NMR Spectra in Liquid and in Solid State



# Spectral Resolution Enhancement in Solid State NMR



dipol-dipol coupling:



magic angle spinning:

$$\hat{\mathbf{R}}_{2,0} \rightarrow 0$$

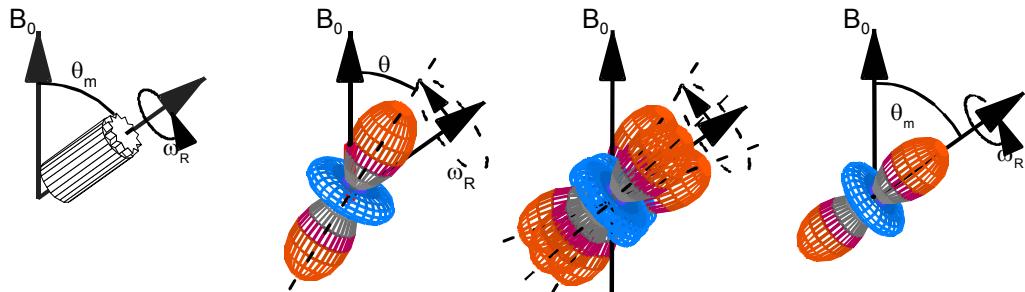
$$\hat{H} = \hat{\mathbf{R}}_{2,0} \cdot \hat{\mathbf{T}}_{2,0}$$

space

$$\hat{H} \propto \frac{1}{r_{ij}^3} \frac{1}{2} (3 \cos^2 \theta_{ij} - 1)$$

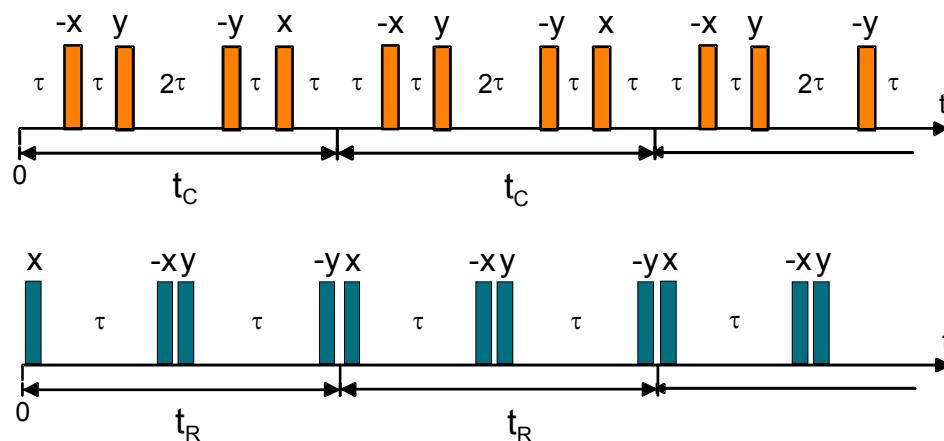
spin

$$\gamma_i \gamma_j (3 \hat{I}_{Z,i} \hat{I}_{Z,j} - \hat{I}_i \cdot \hat{I}_j)$$



RF irradiation:

$$\begin{aligned} \hat{\mathbf{T}}_{2,0} &\rightarrow 0 & (\text{CRAMPS}) \\ \hat{\mathbf{T}}_{2,0} &\rightarrow \text{H}_{D,\text{eff.}} & (\text{Recoupling}) \end{aligned}$$



# Double Quantum NMR Spectroscopy under MAS



preparation

evolution  $t_1$

reconversion

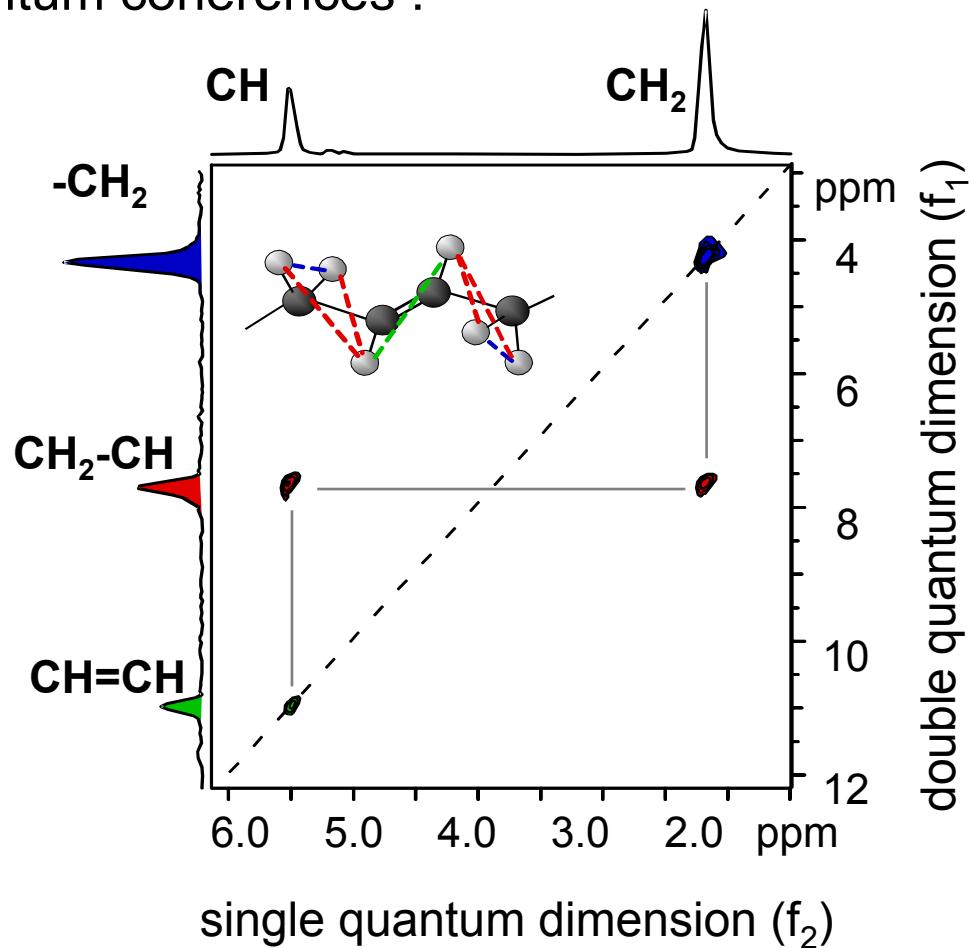
detection  $t_2$

properties of double quantum coherences :

$$\omega_{DQ} = \sum_i \omega_{SQ,i}$$

$$I_{DQ,ij} = f(D_{ij} \cdot t)$$

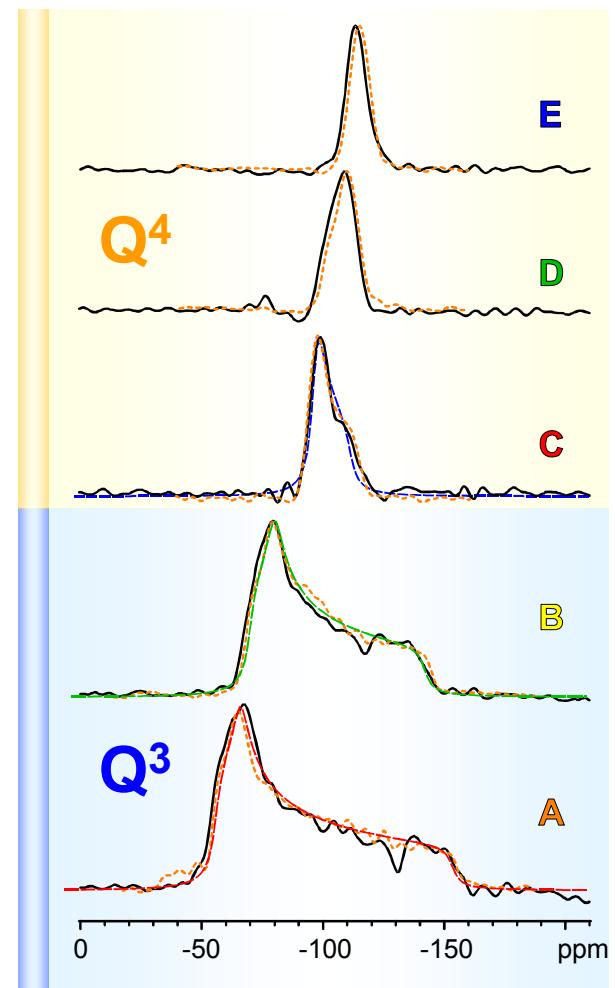
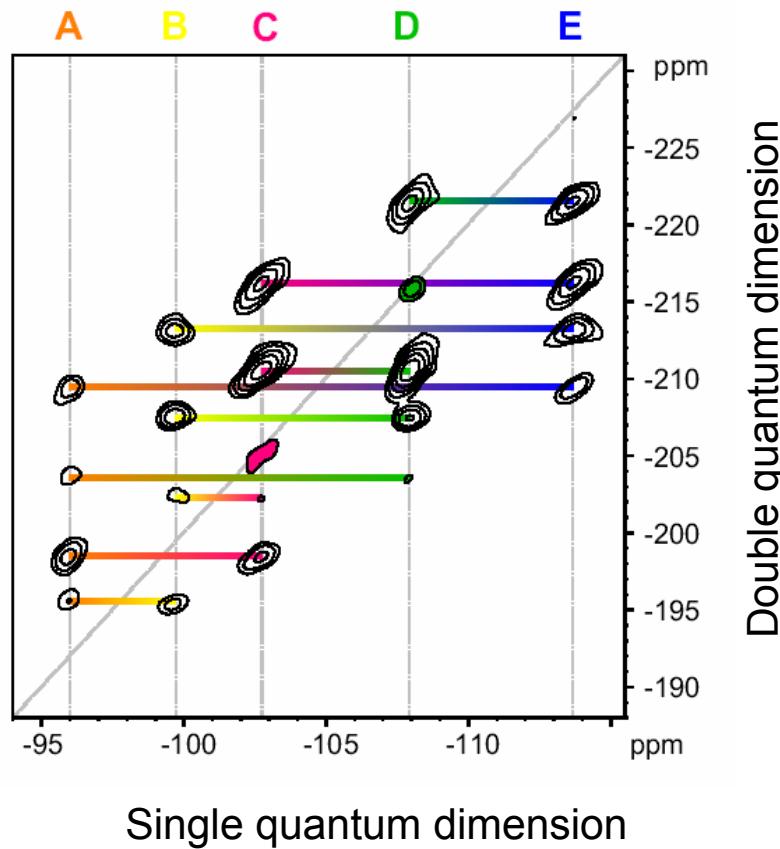
$$\frac{dM}{dt} \approx 0$$



# Molekulare Stuktur von Silikat-Schichten



$^{29}\text{Si}$  double quantum spectrum: DQ intensity  $\propto r^{-6} \Rightarrow$  coordination shells

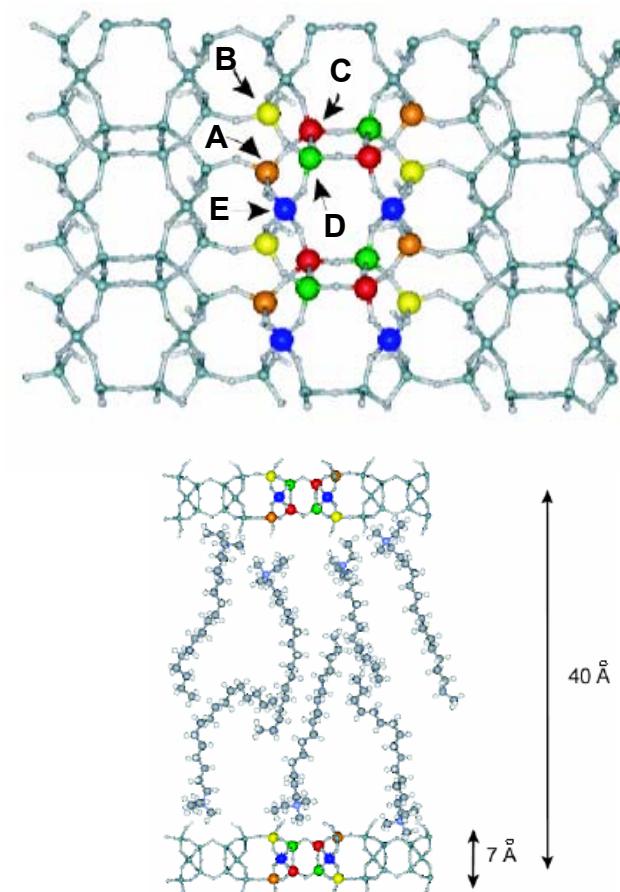
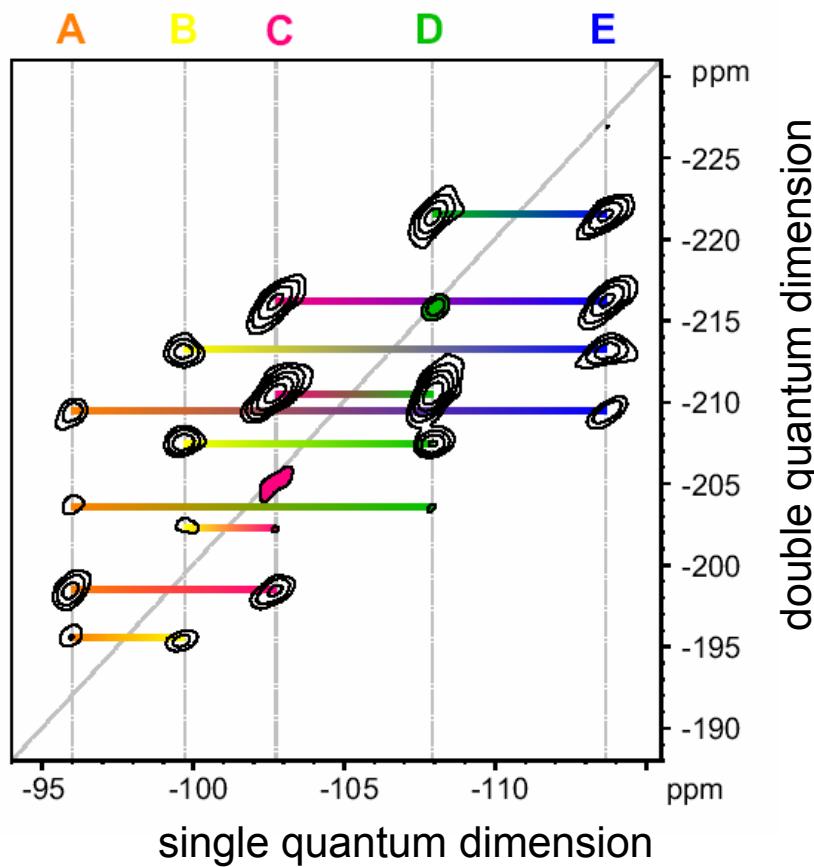


Analysis of layered silicates via next nearest neighbor relations

# Molecular Structure of Layered Silicates



$^{29}\text{Si}$  double-quantum spectrum: DQ-Intensities => coordination spheres



structure of layered silicates from analysis of spatial proximities

N. Hedin et al., *J. Am. Chem. Soc.* **126**, 9425 (2004).

# Polybenzoxazines

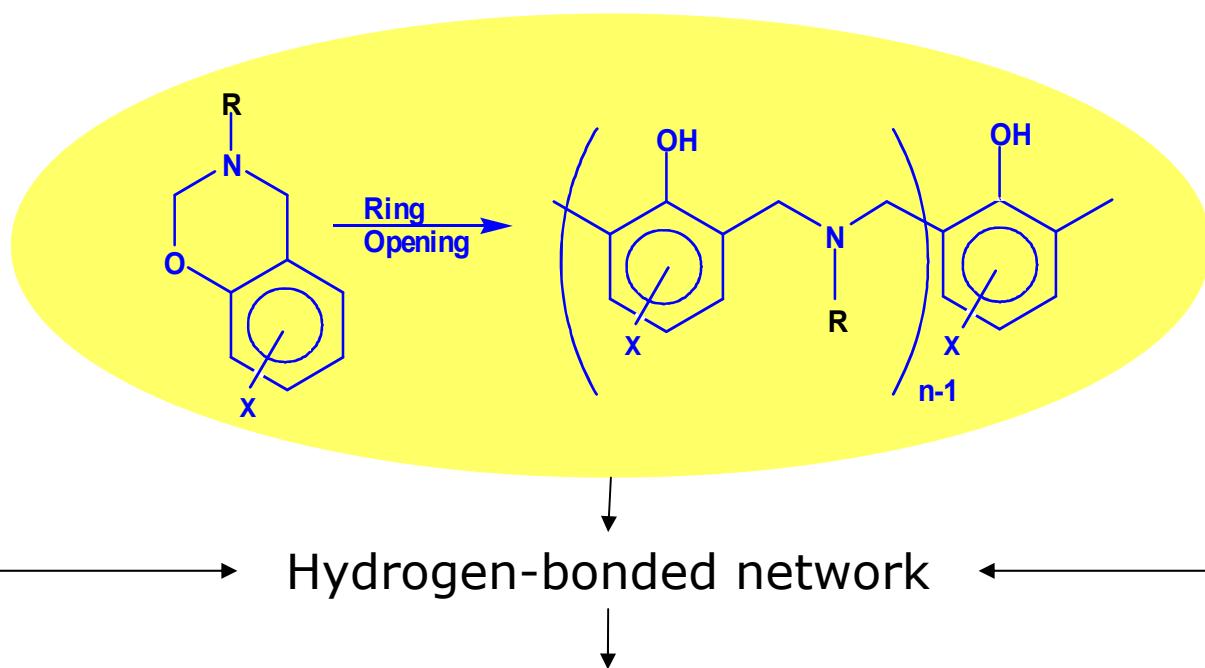


## Useful Properties

- High  $T_g$
- Good mechanical properties
- Excellent UV and chemical resistance

## Unusual Properties

- Low water absorption
- Low volumetric expansion on curing
- High modulus



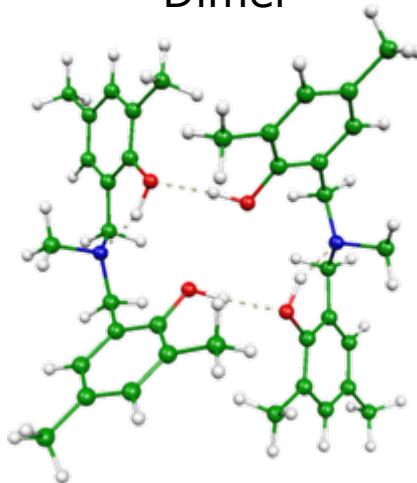
What is the nature of the network?



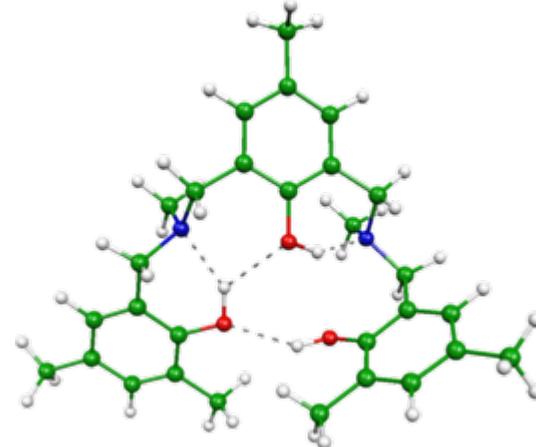
# Benzoxazine Oligomers Studied by $^1\text{H}$ DQ NMR



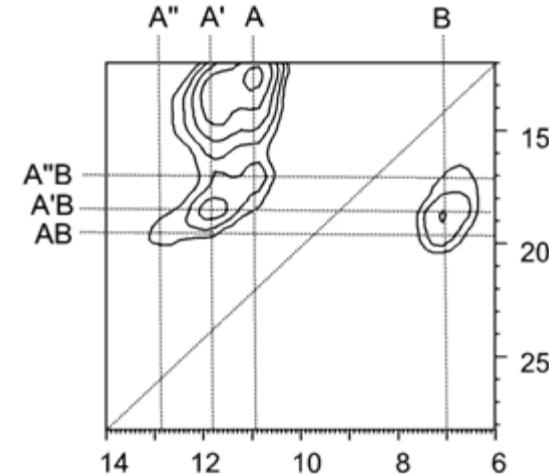
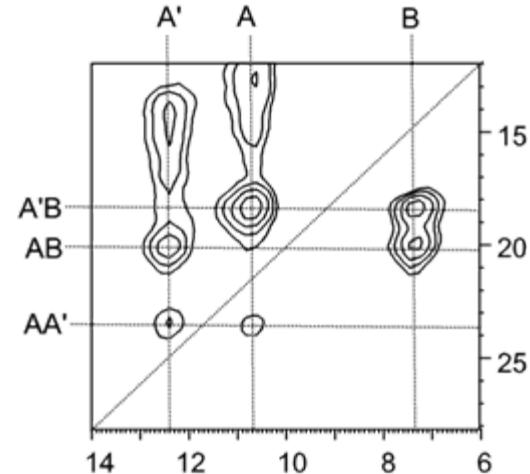
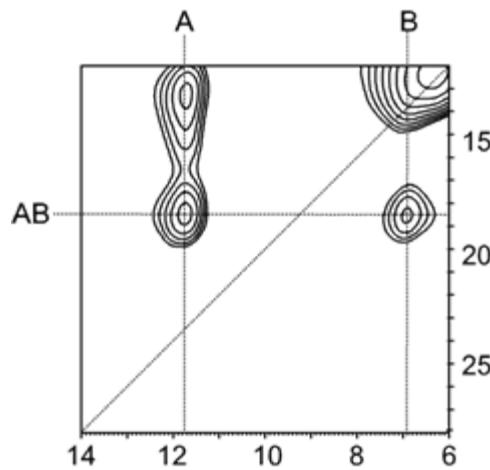
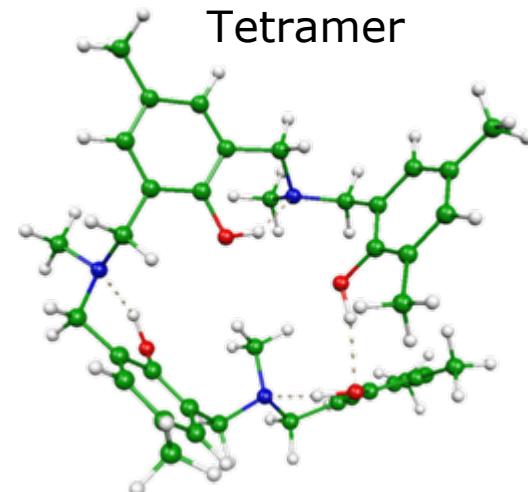
Dimer



Trimer



Tetramer

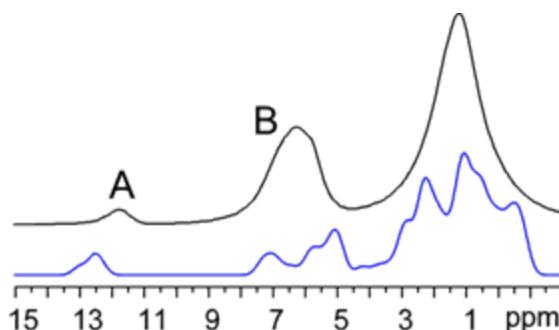


Changes in hydrogen bonding structure evident from changing  $^1\text{H}$  resonances and DQ contacts

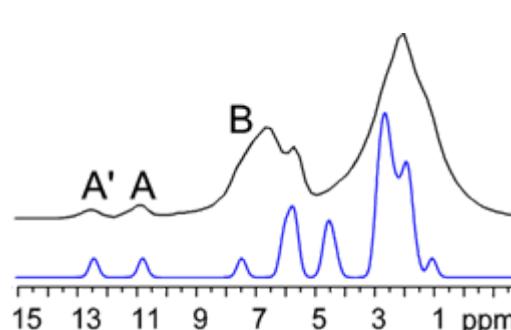
# Hydrogen Bonds Assigned via DFT-Based Chemical Shift Calculations



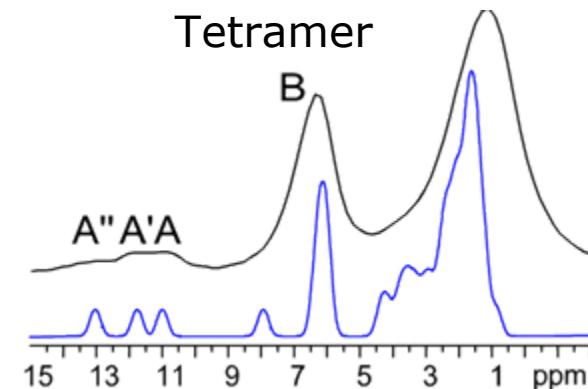
Dimer



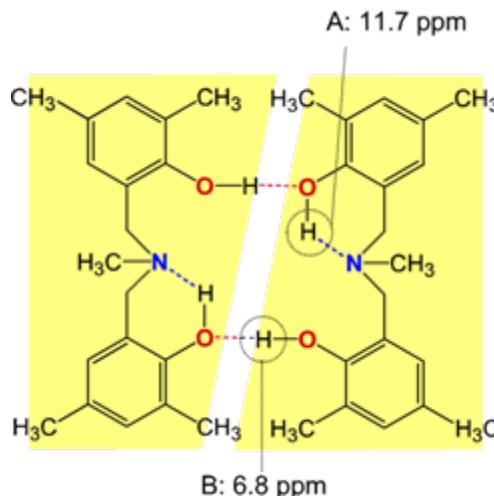
Trimer



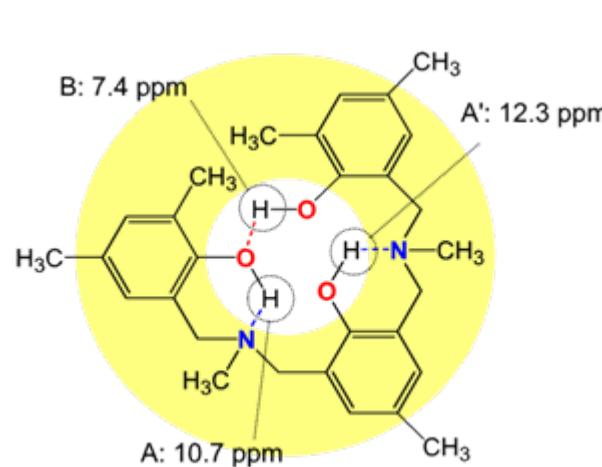
Tetramer



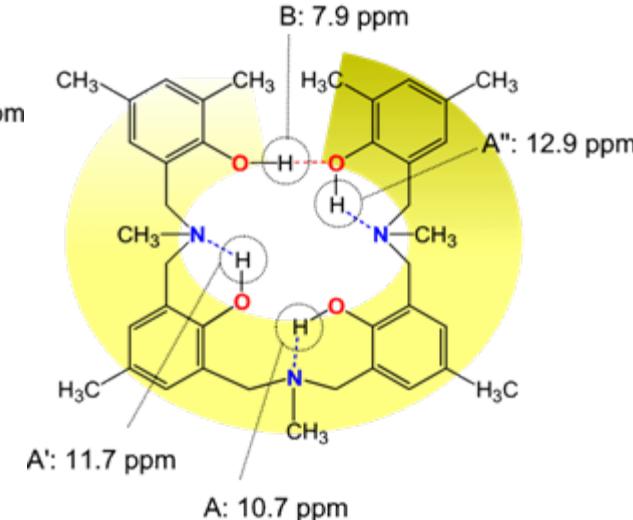
**Calculated** and **experimental**  $^1\text{H}$  NMR spectra agree, particularly in hydrogen-bonding region.



Isolated Dimer Pair - X-ray Structure  
Intra:Inter = 1:1



Trimer forms Planar Ring  
Intra:Inter = 2:1

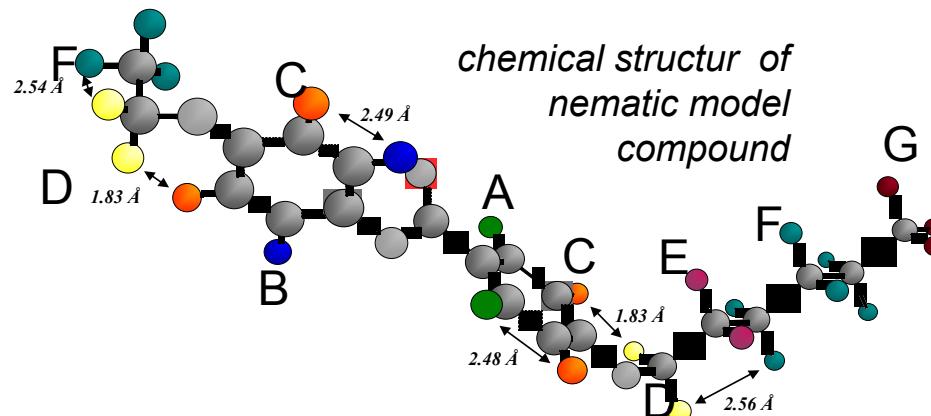


Tetramer forms Overlapping Loop  
Intra:Inter = 3:1



$^1\text{H}$  chemical shifts influenced by proximity to oxygen. Helical structure predicted for polymer.

# Order Parameter in Liquid Crystalline Phases

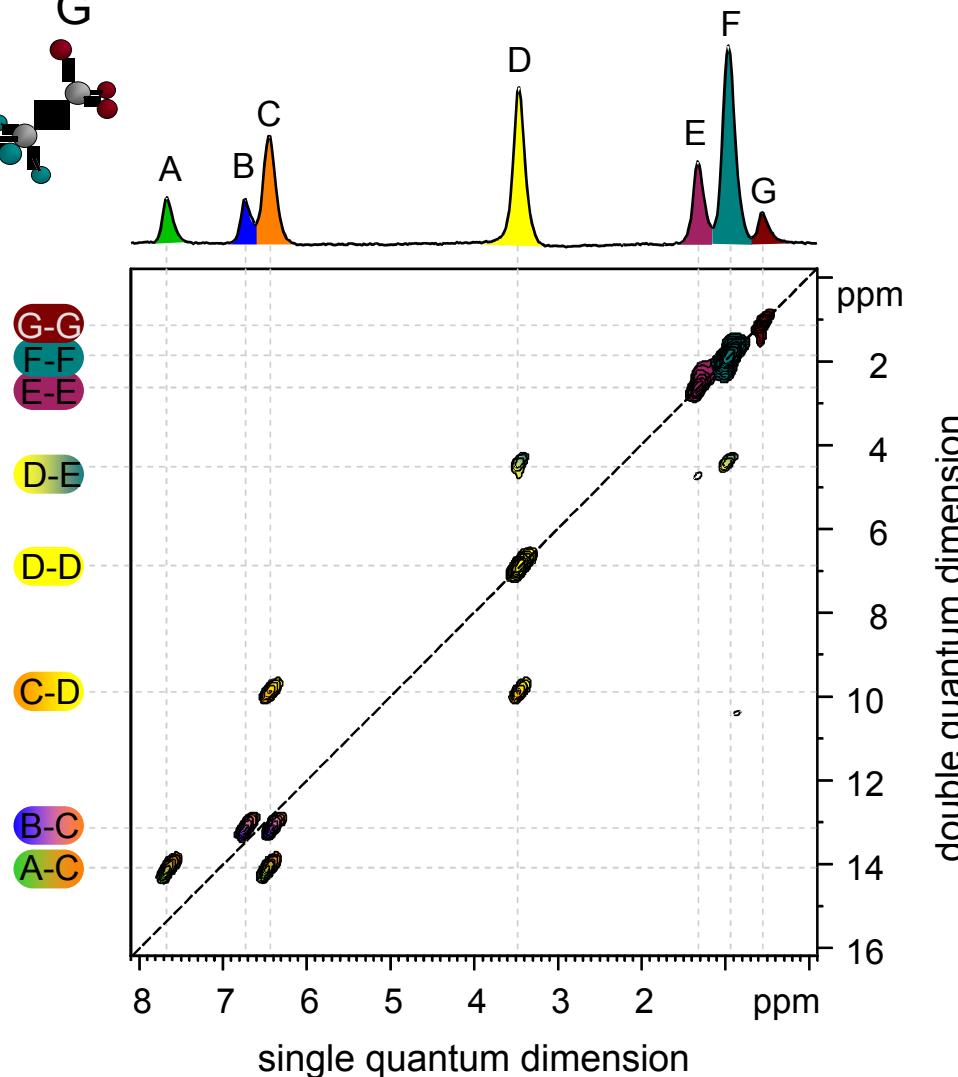


order parameter  $S_{ij}$ :

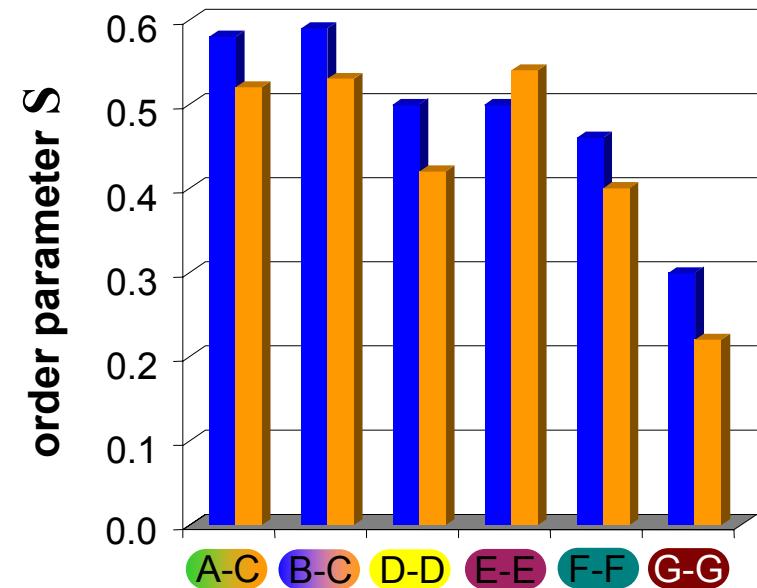
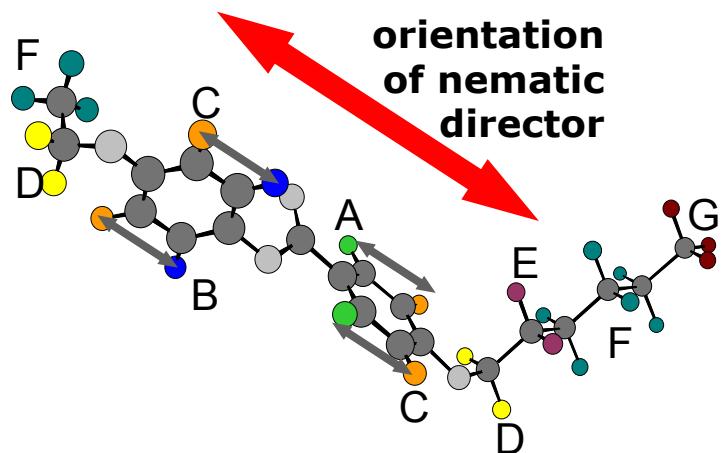
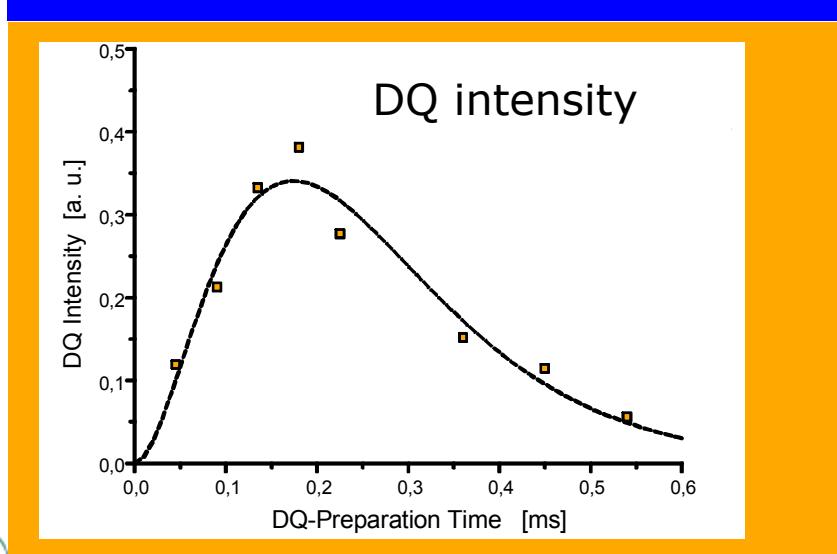
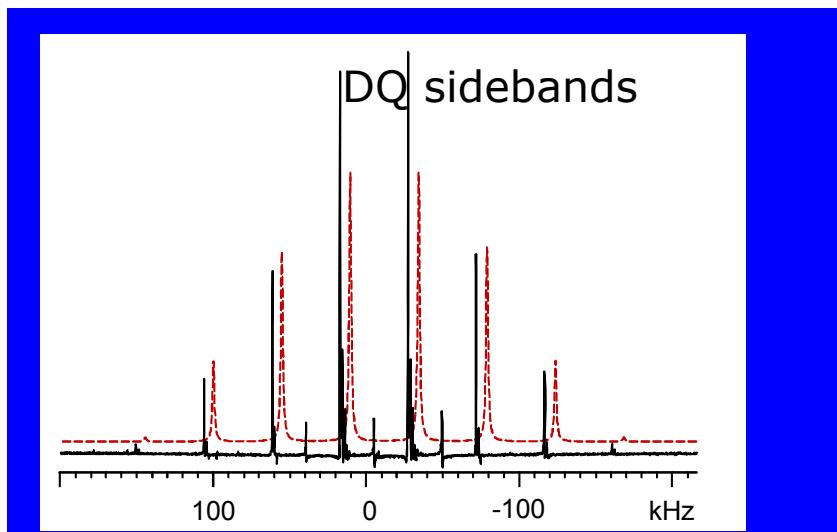
$$S_{ij} = \left\langle \frac{1}{2}(3 \cos^2 \theta - 1) \right\rangle$$



$$S_{ij} = \frac{\left\langle D_{ij,\text{eff}} \right\rangle}{D_{ij,\text{stat}}}$$



# Order Parameters in Liquid Crystalline Systems





# **Investigation of Structure and Dynamics in Polymeric Systems via Solid State NMR**

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**Introduction** • Interaction in solid state NMR

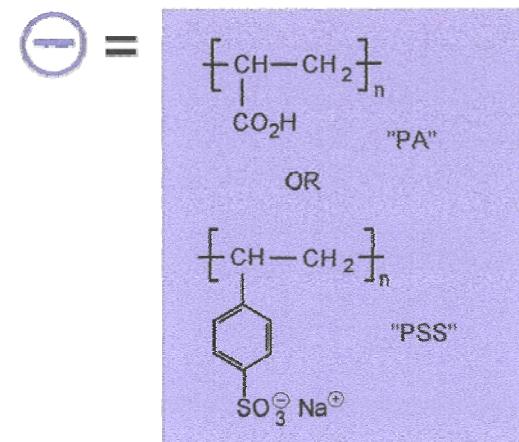
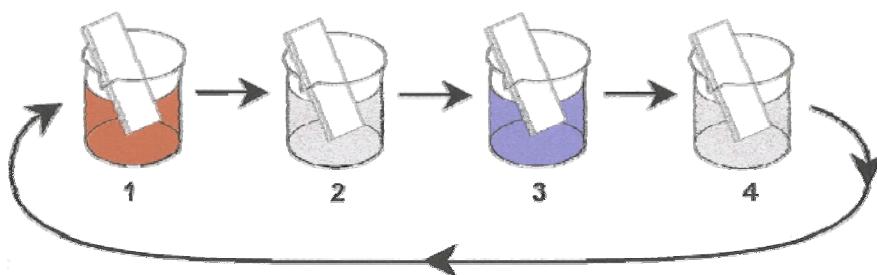
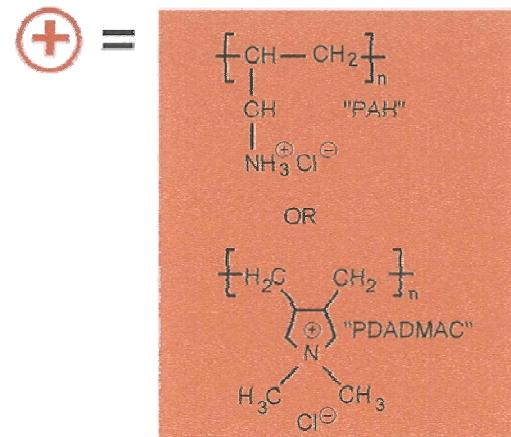
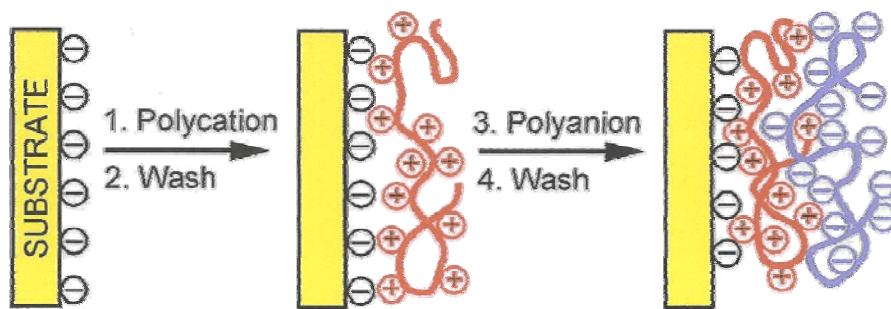
**Solid State NMR** • MAS, recoupling, double-quantum NMR

**Polymer Dynamics** • Polyelectrolyte multi layers, reptations-model, scaling laws in polymer dynamics, influence of rigid confinements, conformational stability in PEMA melts.

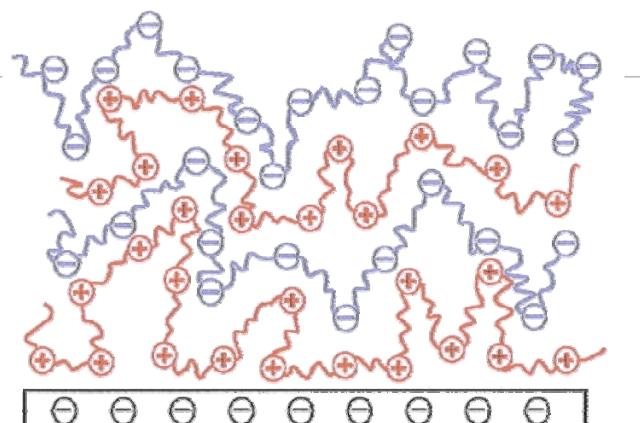
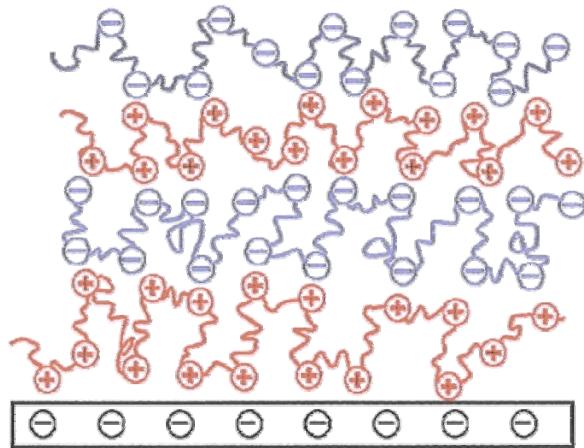
**Conclusions** • Pro and Contra Solid State NMR investigations



# Polyelectrolyte Multi-Layers

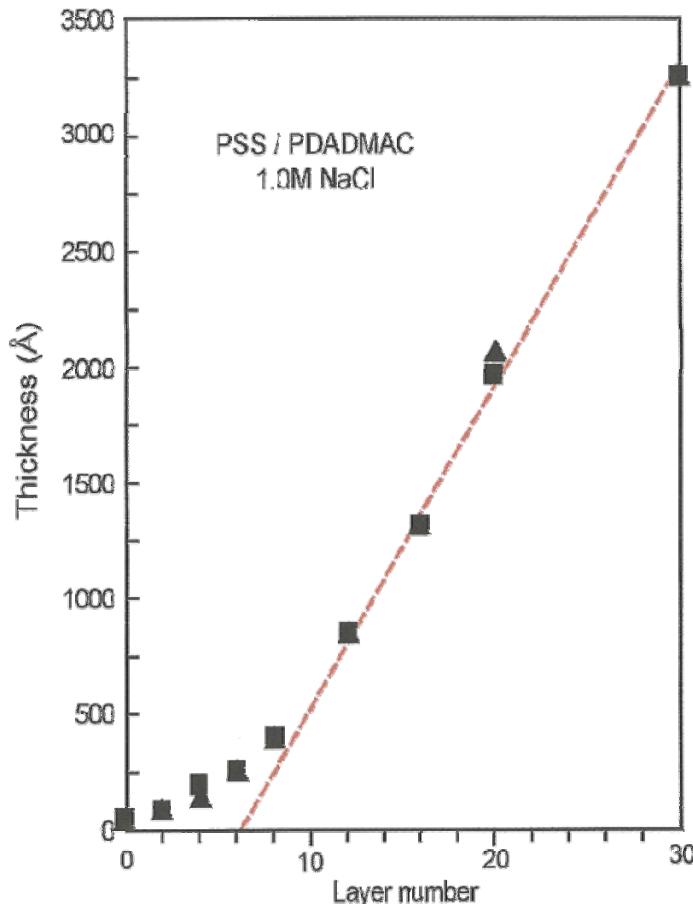


# Structure of Polyelectrolyte Layers



interdigitated but still stratified chains

Observed Film Growth ( Ellipsometry)



# DQ NMR Investigation of Structure: PEM vs. PEC



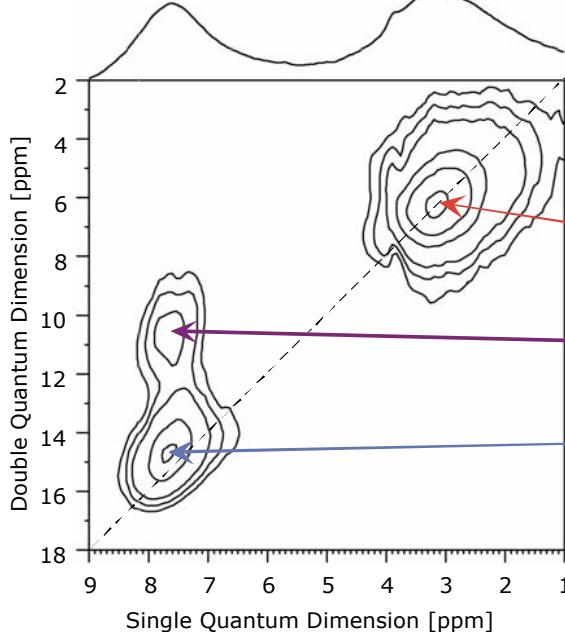
## Polyelectrolyte Complex

Polycation Solution

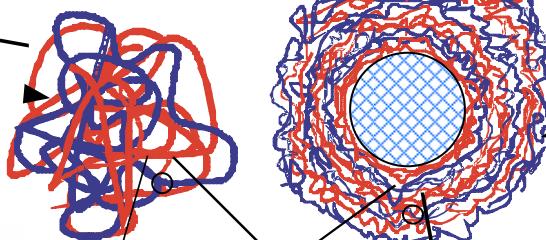
Polyanion Solution

After Drying

Polyelectrolyte Complex Formation



After 3 cycles

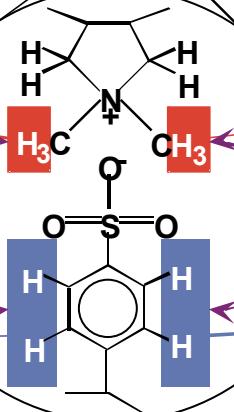
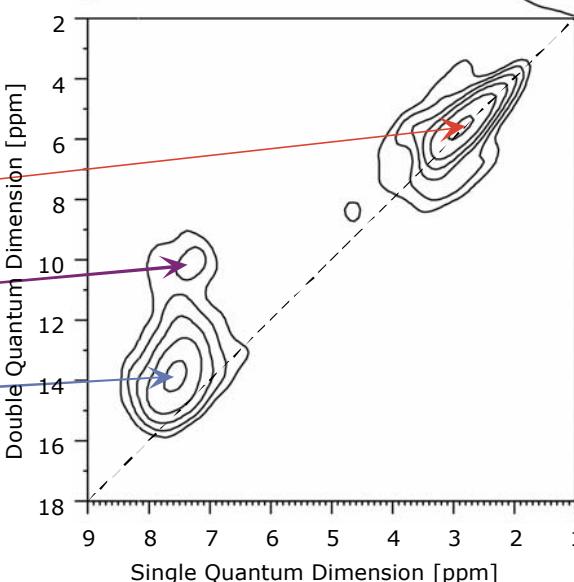


## Polyelectrolyte Multilayer

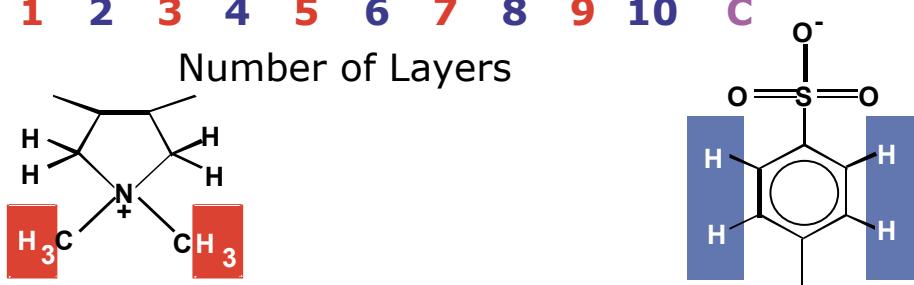
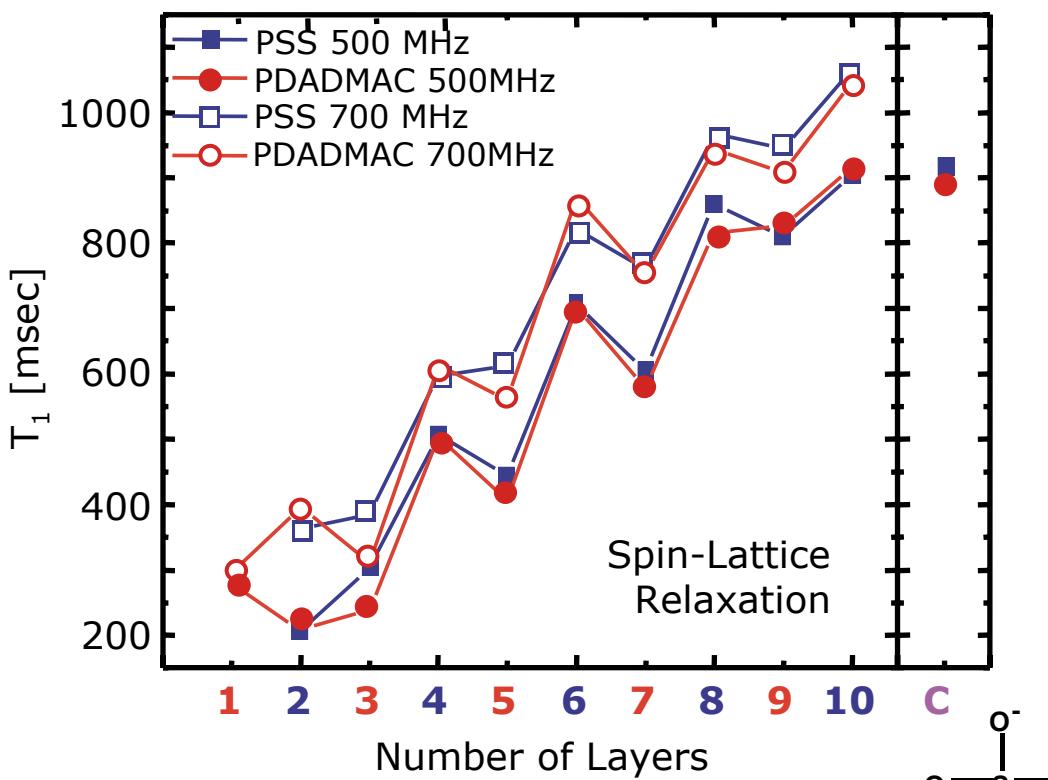
Anionic Substrate  
Polycation Solution

Aqueous Rinse

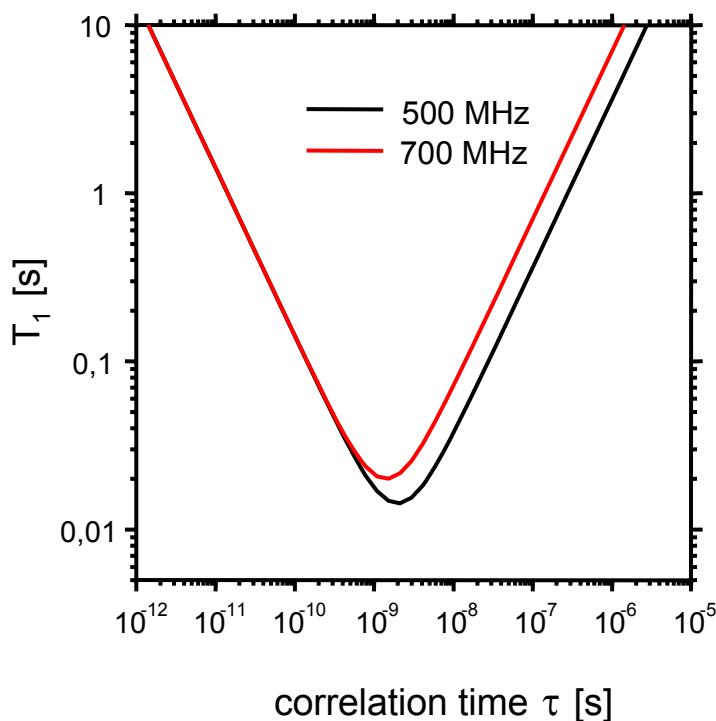
Polyanion Solution



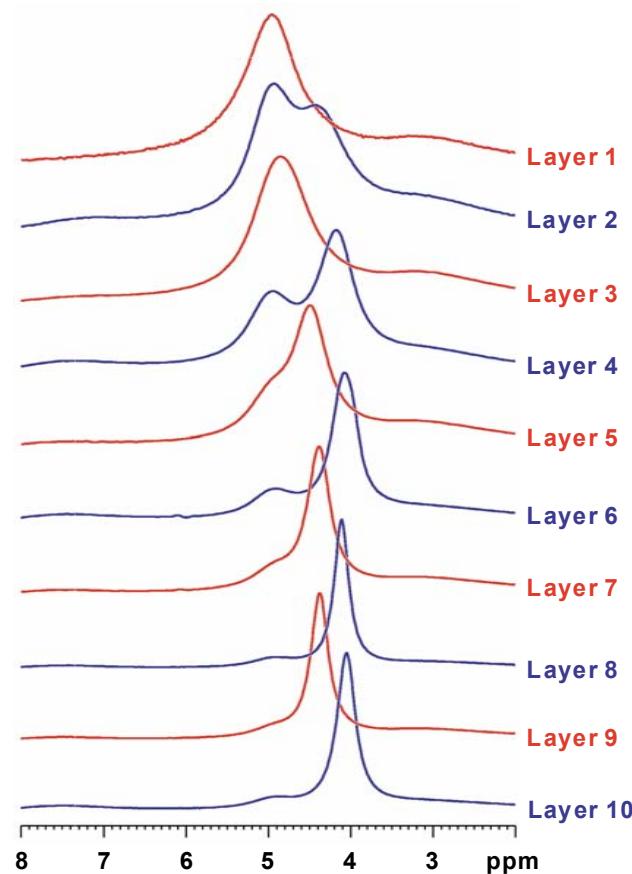
# Molecular Dynamics in Polyelectrolyte Multi Layers



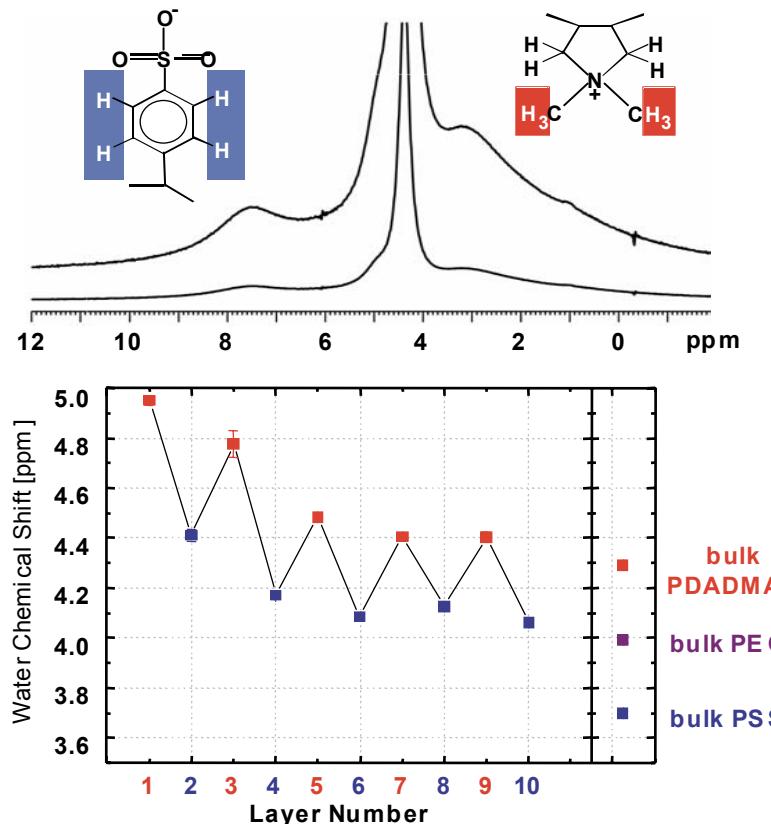
theoretical  $T_1$  relaxation behavior:



# Localization of Water in Polyelectrolyte Multilayers



$^1\text{H}$  MAS NMR Spectra



$^1\text{H}$  Chemical Shift

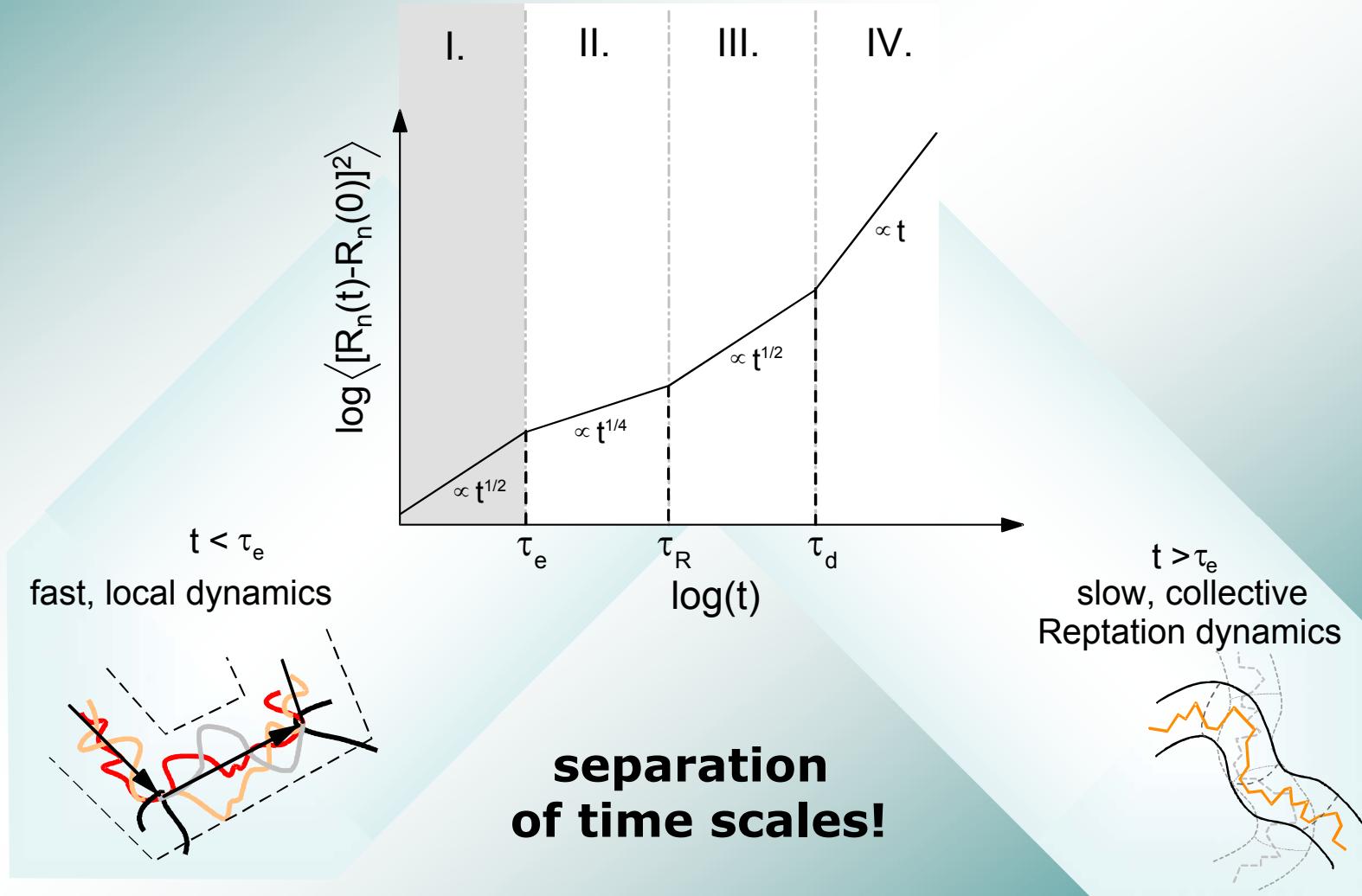


M. McCormick et al., *Macromolecules* **36**, 3616 (2003).  
S. Pawsey et al., *J. Am Chem. Soc.* **125**, 4174 (2003).

# Length- and Time Scales in Polymer Dynamics



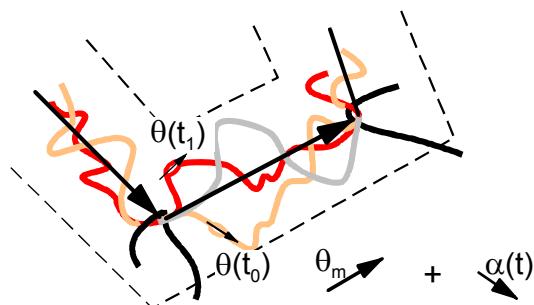
dynamic regimes of the Reptation model :



# DQ Measurements of Dynamics on Different Time Scales



$$I_{DQ} \propto \left\langle \int_0^t dt' \int_{t+t_1}^{2t+t_1} dt'' \left\langle D_{ij,\text{eff}} \right\rangle^2 \cdot d_{2,-m}^{(2)}(t') d_{2,m}^{(2)}(t'') \right\rangle$$



local order parameter :

$$S_{ij} = \left\langle D_{ij,\text{eff}} \right\rangle / D_{ij}$$

static systems :

$$S_{ij}(t)=1$$

isotropic motion :

$$S_{ij}(t)=0$$

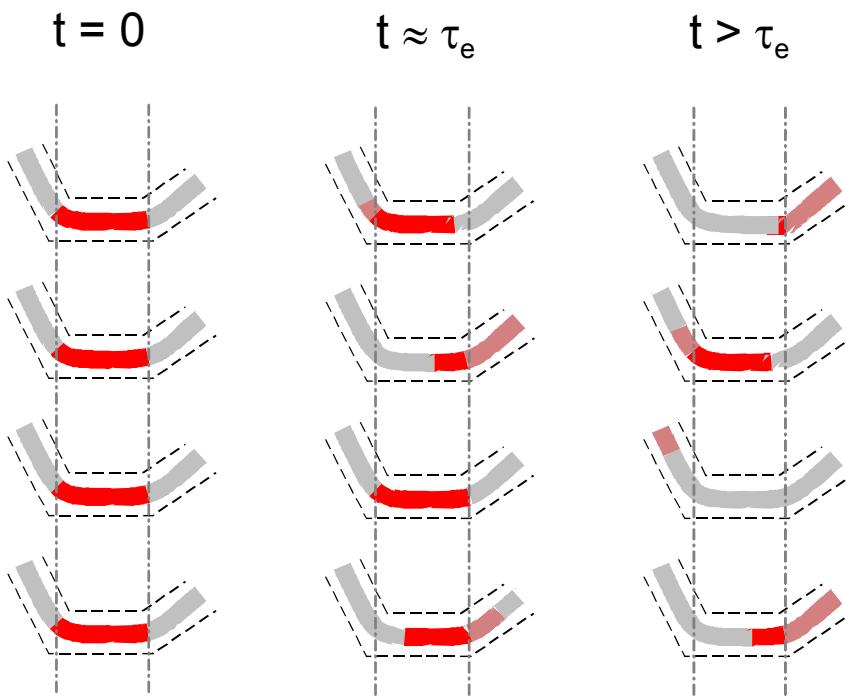
polymer network theory :

$$S \approx \frac{3}{5} N_e^{-1}$$



Polybutadien :

$$S \approx \frac{3}{5} \frac{M_{\text{Kuhn}}}{M_e} \approx 0.03$$

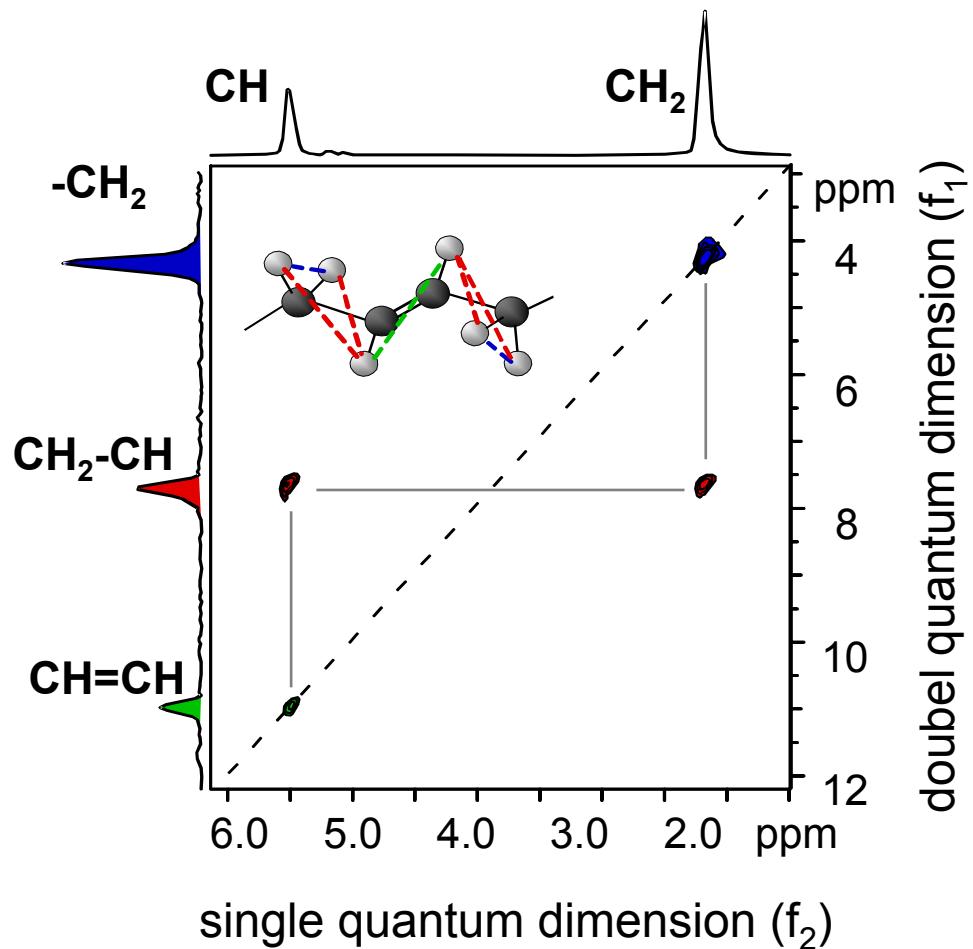


$\left\langle d_{2,-m}^{(2)}(t'_{\text{exc.}}) \cdot d_{2,m}^{(2)}(t''_{\text{rec.}}) \right\rangle_t$  corresponds to  
**return-to-origin** probability  $C(t)$

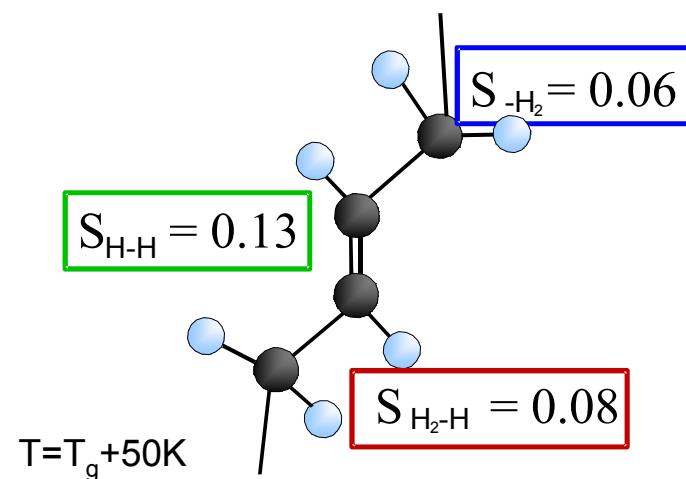
# Local Order Parameters in 1,4 Polybutadien Melts



$^1\text{H}$  double quantum NMR spectrum



Dynamic order parameter  $S$   
via residual dipolar couplings



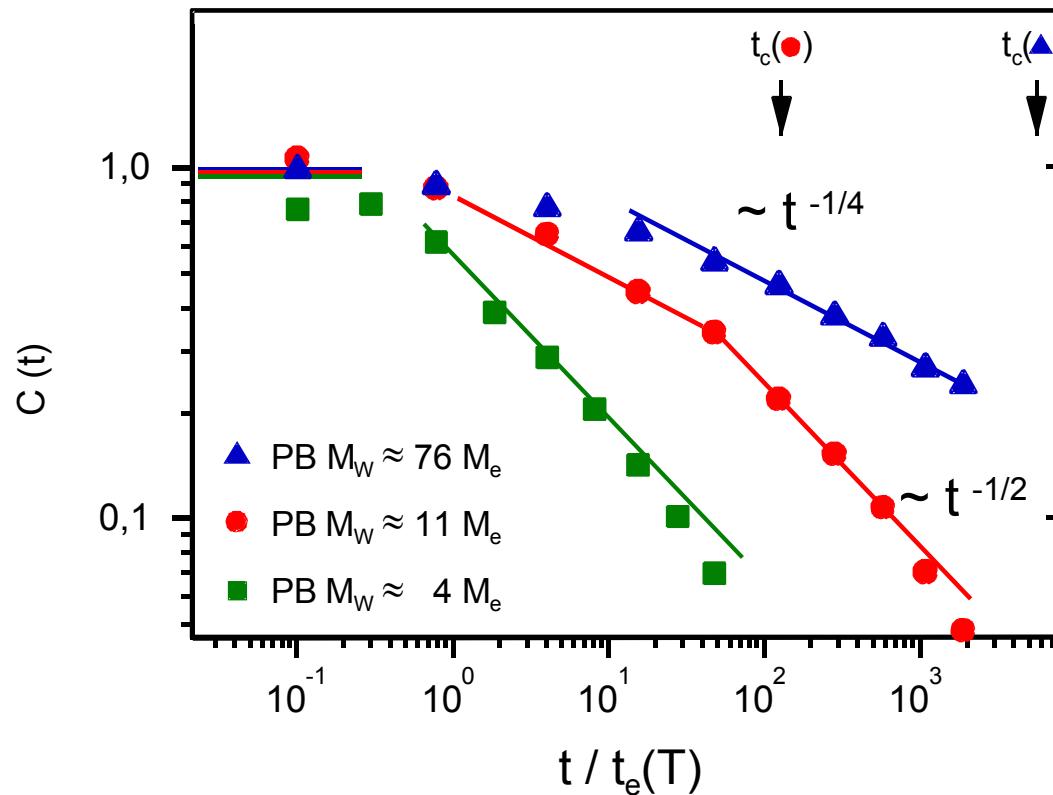
$$\Rightarrow S_{\text{C}=\text{C}} = 0.20 \pm 0.05$$



# Time Dependence of Local Order Parameter



double-quantum filtered experiments on 1,4 poly-butadien

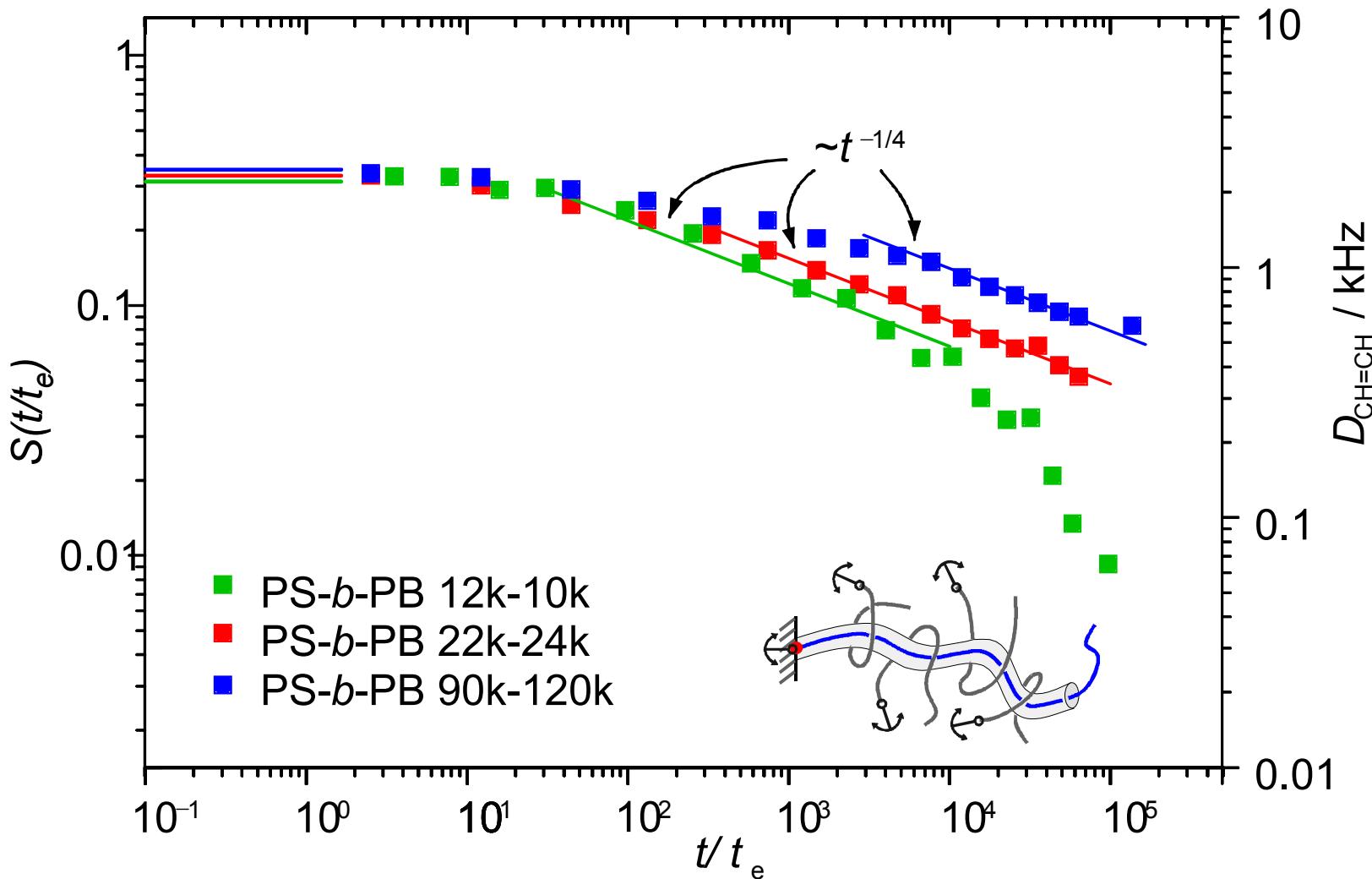


Reptation-model predicts two scaling laws:

$$S \sim t^{-1/4} \text{ and } S \sim t^{-1/2}$$

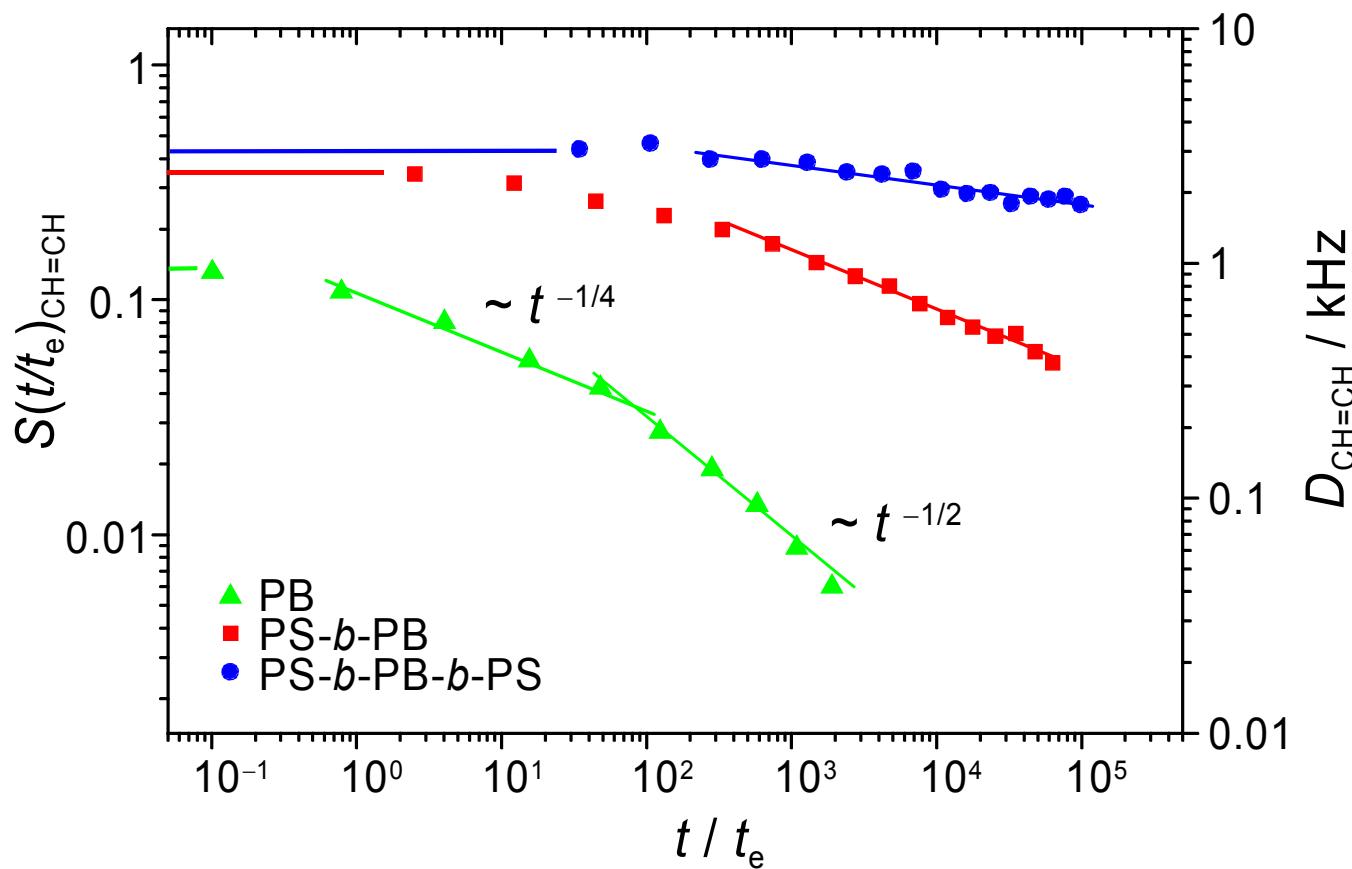


# Molecular Weight Dependent Dynamics of PB Melts in PS-PB



Tethering a PB chain end to a rigid PS block stabilizes the  $t^{-1/4}$ -regime

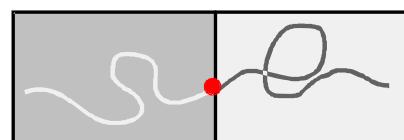
# Influence of Rigid Confinements on Polymer Dynamics



PB



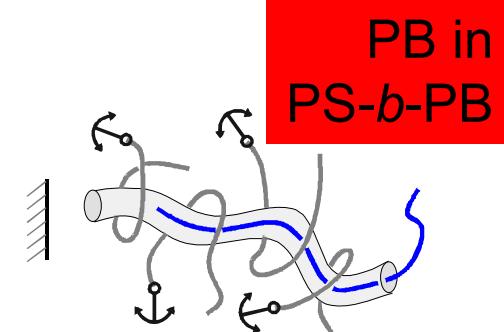
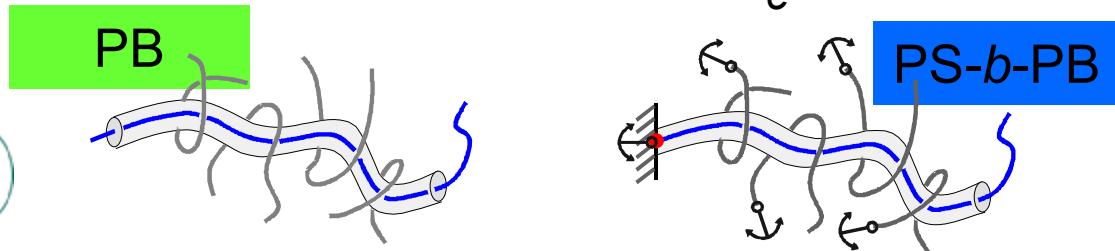
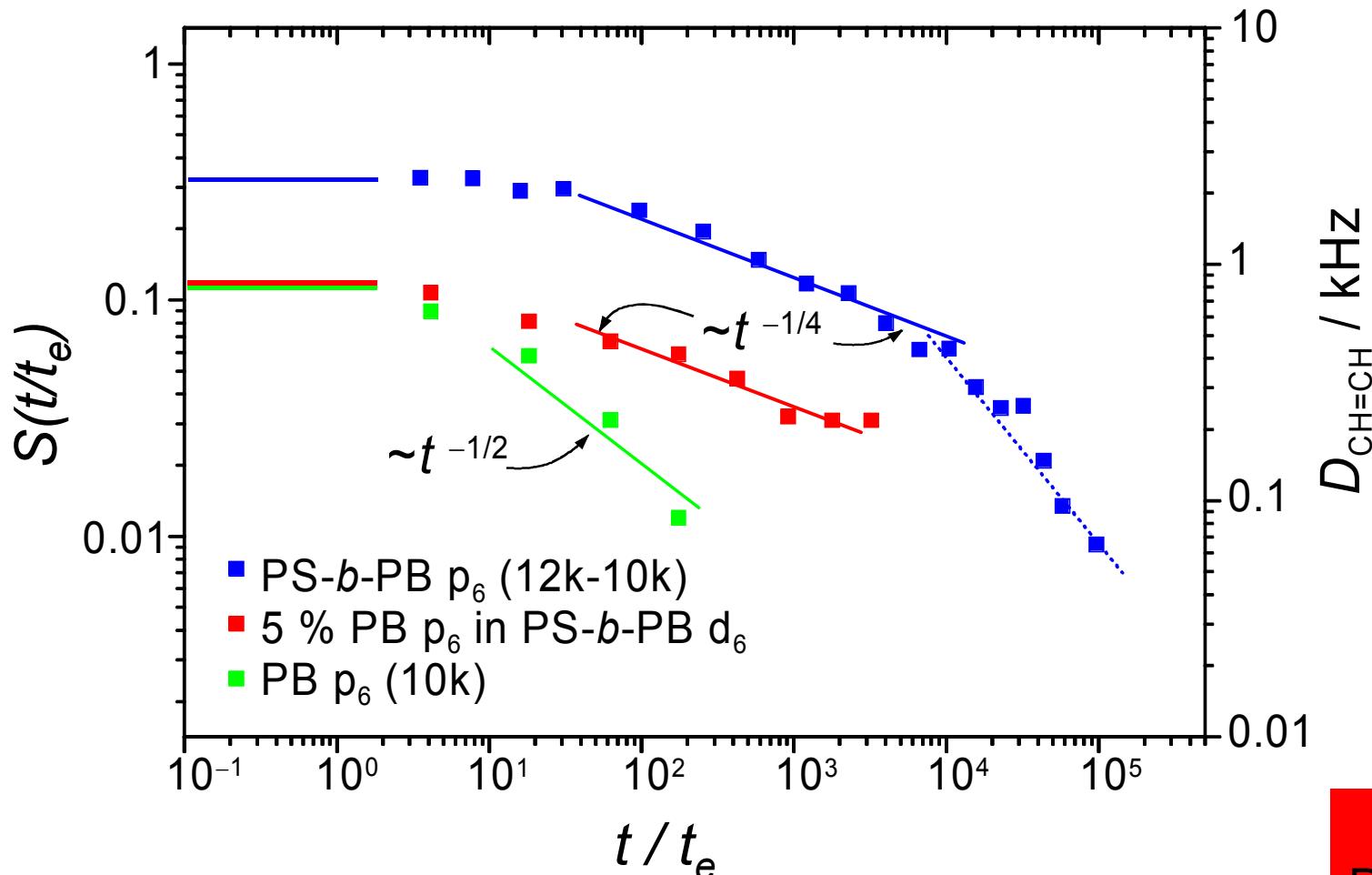
PS-*b*-PB



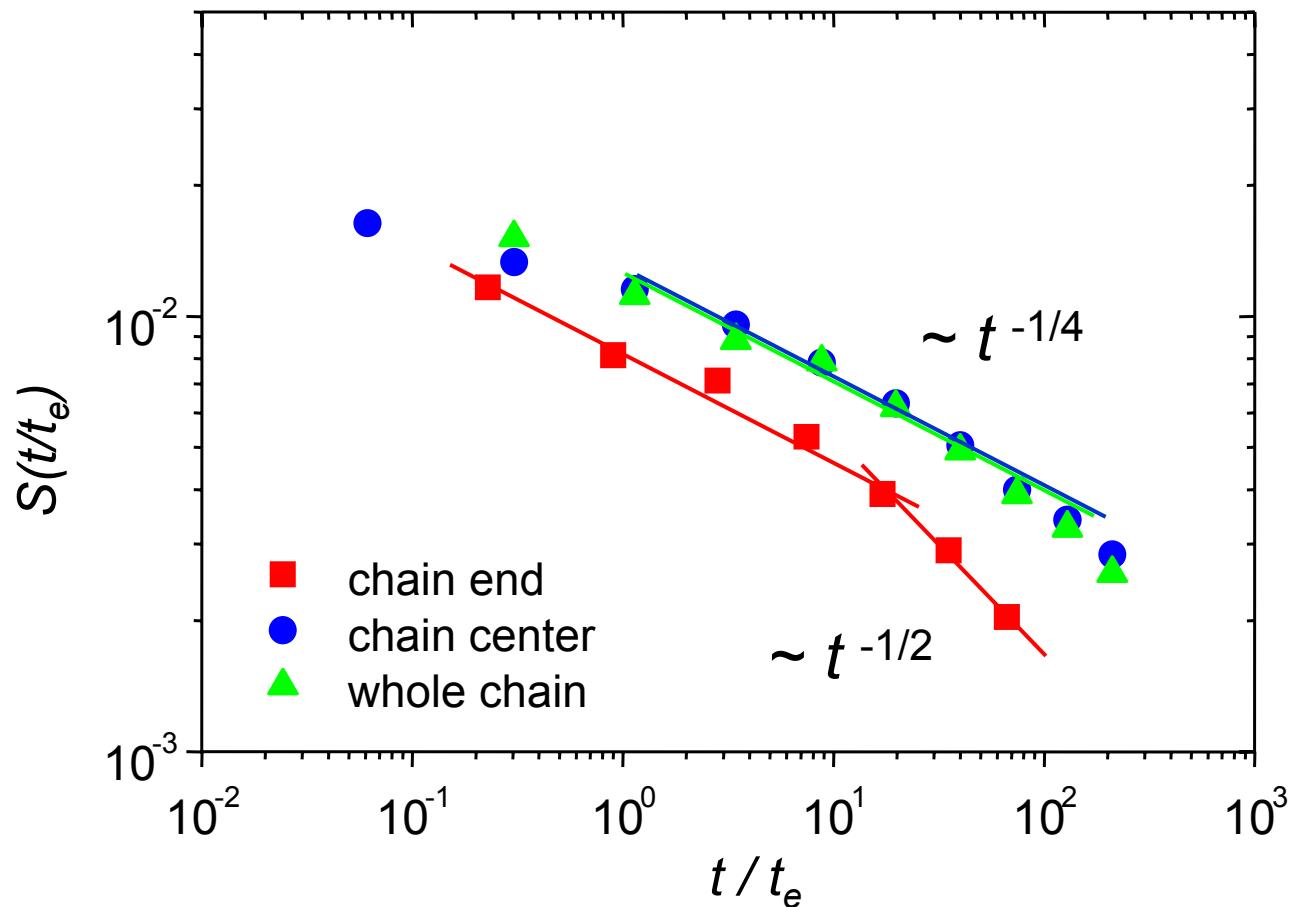
PS-*b*-PB-*b*-PS



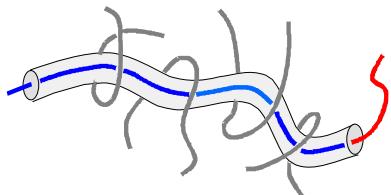
# Polymer Dynamics in heterogeneous Polymer Melts



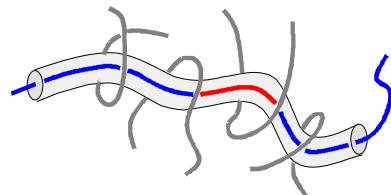
# Variation of Dynamic Order Along the Polymer Chain



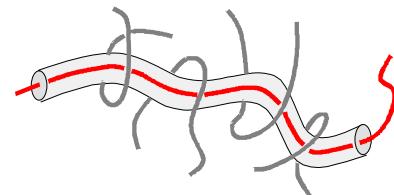
PB( $d_6$ )-PB



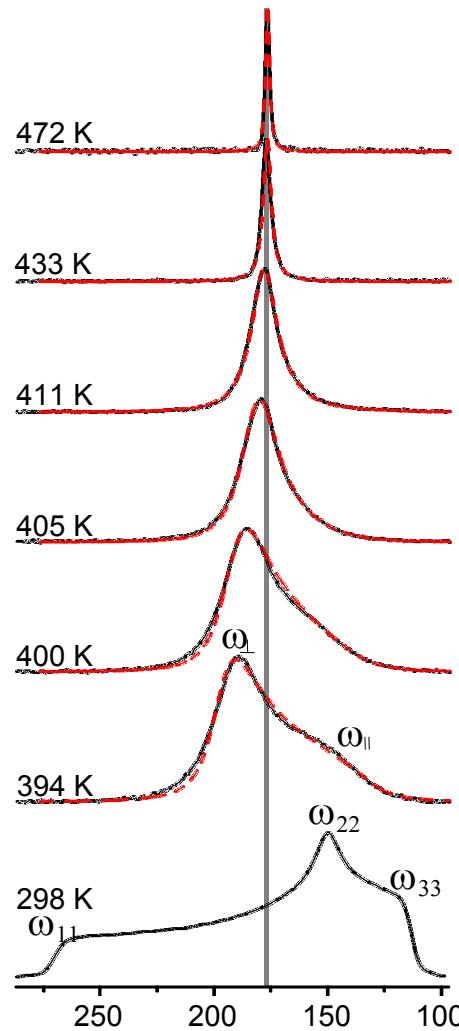
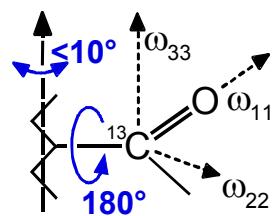
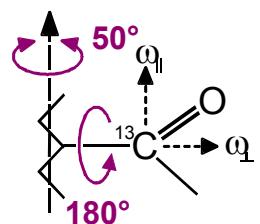
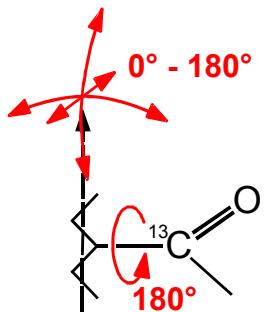
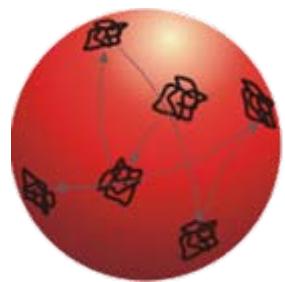
PB( $d_6$ )-PB-PB( $d_6$ )



PB



# **a-PEMA: Isotropisation of Chain Dynamics**



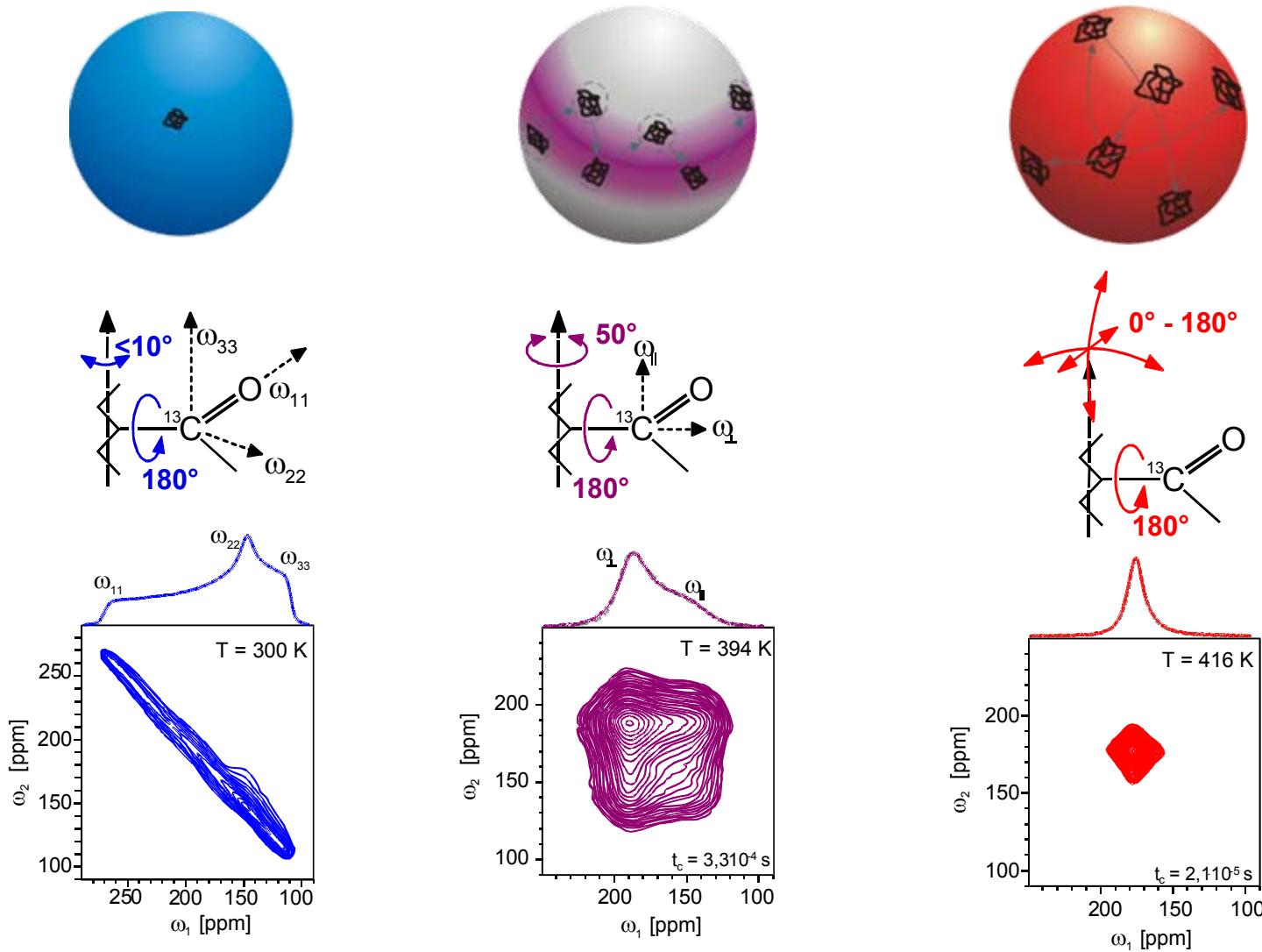
Melt

Melt

$T_g$   
Glass



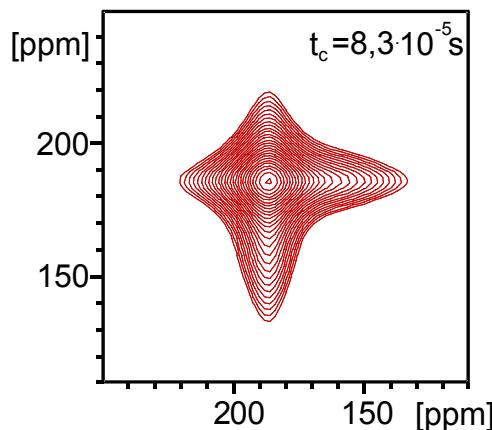
# a-PEMA: Isotropisation of Chain Dynamics



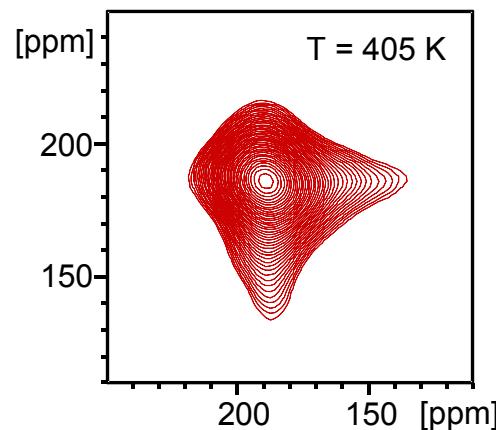
# Dynamic Models: Random Jump vs. Rotational Diffusion



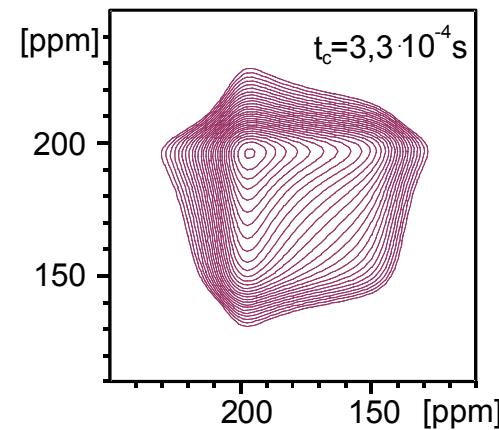
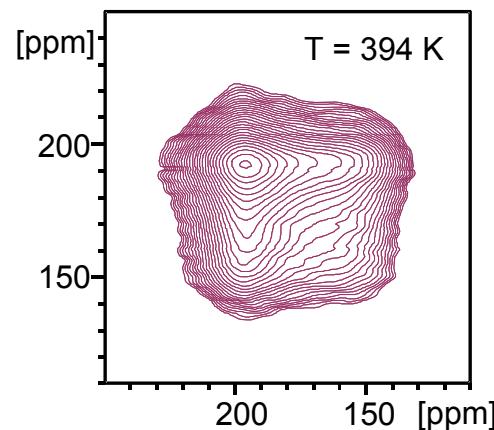
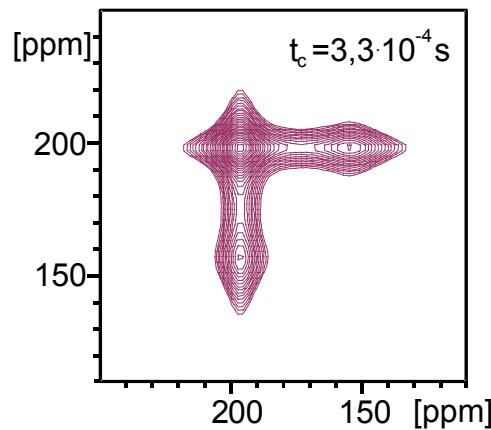
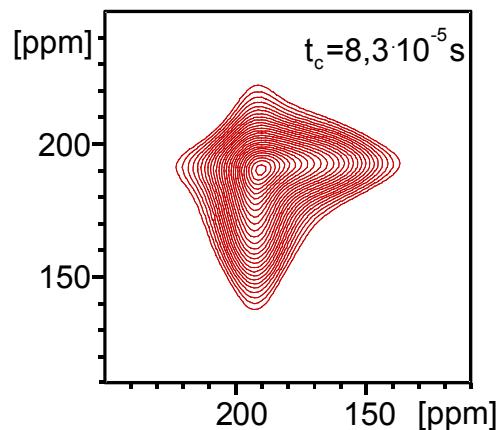
rotational diffusion



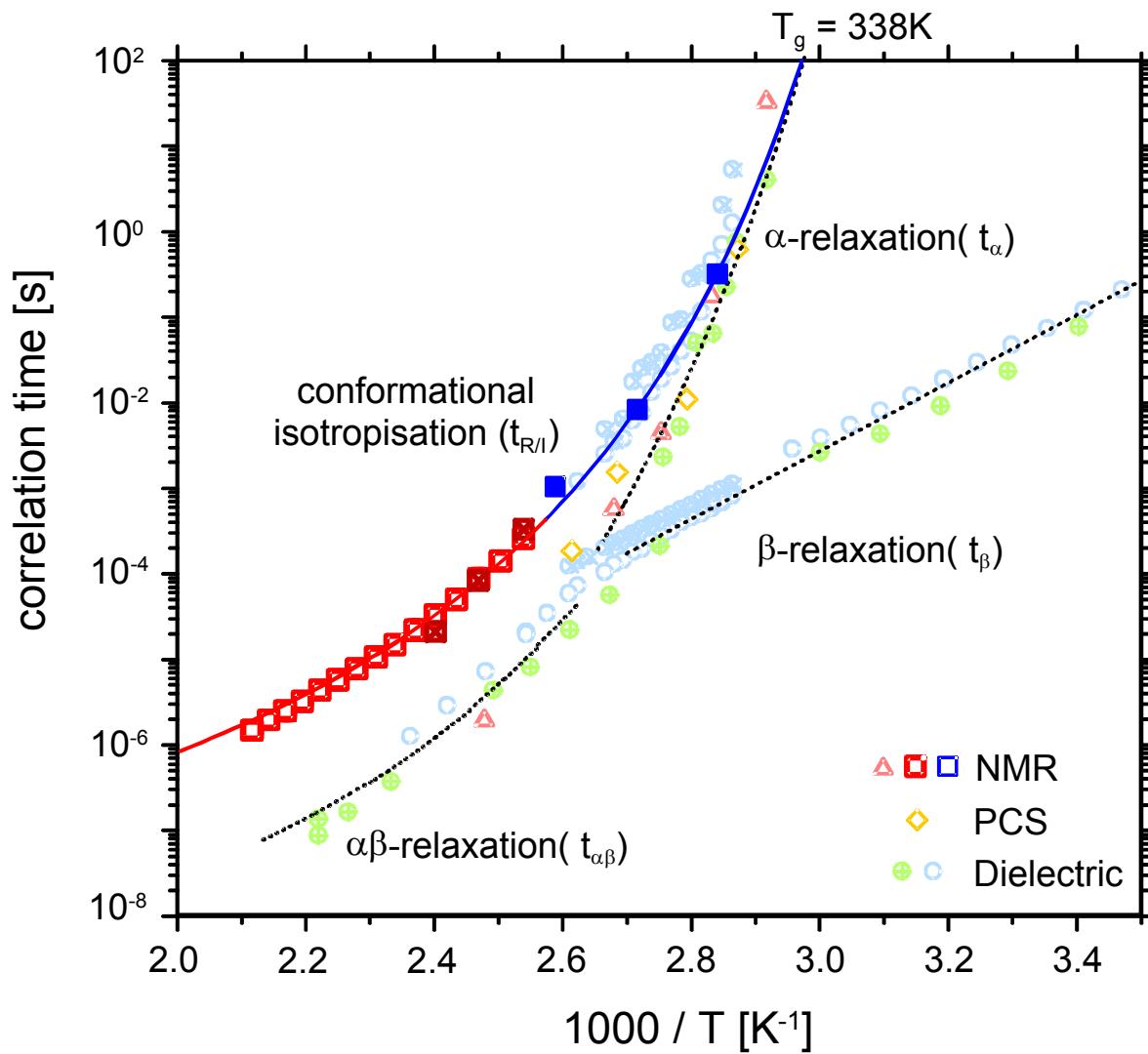
experimental results



random jump



# Time Scales of Molecular Dynamics PEMA Melts



Arrhenius-diagram of dynamic processes in PEMA

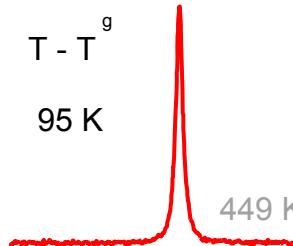
# Length Scale of Isotropisation Process



from radicalic polymerisation

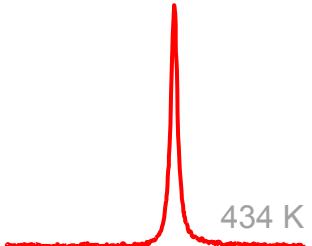
$$M_w = 600 \text{ kg mol}^{-1}$$

$$M_w/M_n = 1,53$$



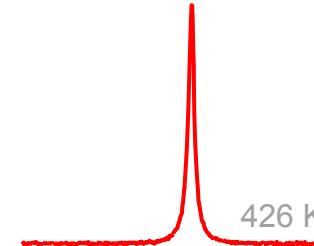
$$M_w = 15,9 \text{ kg mol}^{-1}$$

$$M_w/M_n = 1,61$$



$$M_w = 7,6 \text{ kg mol}^{-1}$$

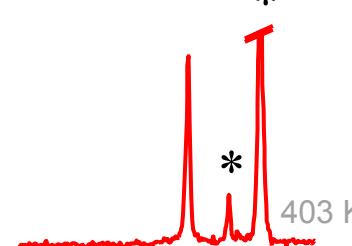
$$M_w/M_n = 1,48$$



from anionic polymerisation

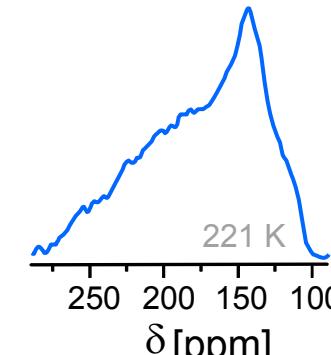
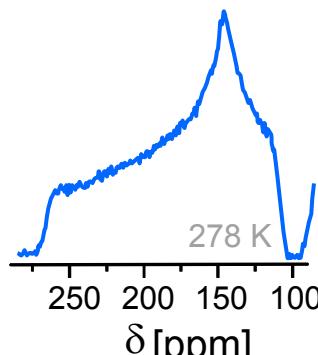
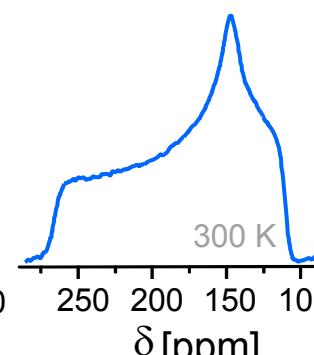
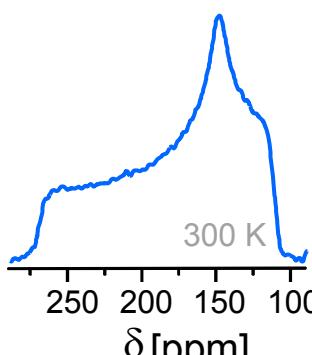
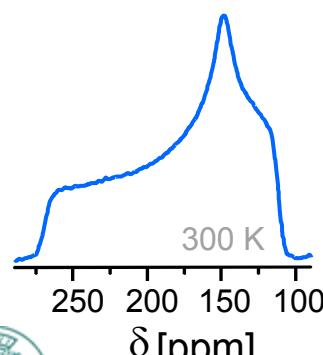
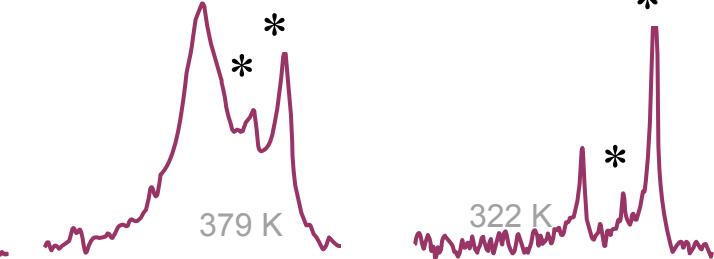
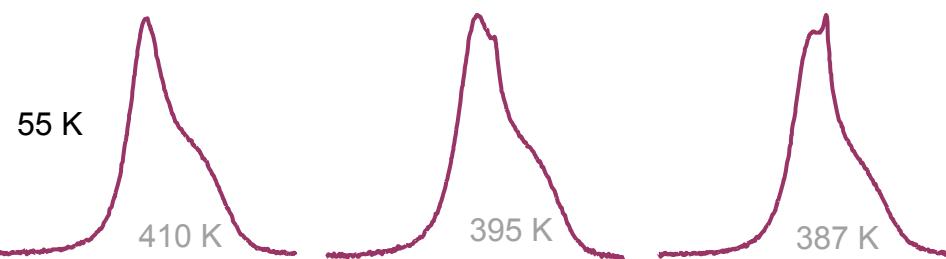
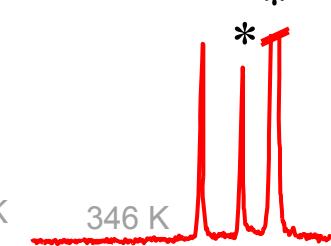
$$M_w = 1,4 \text{ kg mol}^{-1}$$

$$M_w/M_n = 1,07^*$$

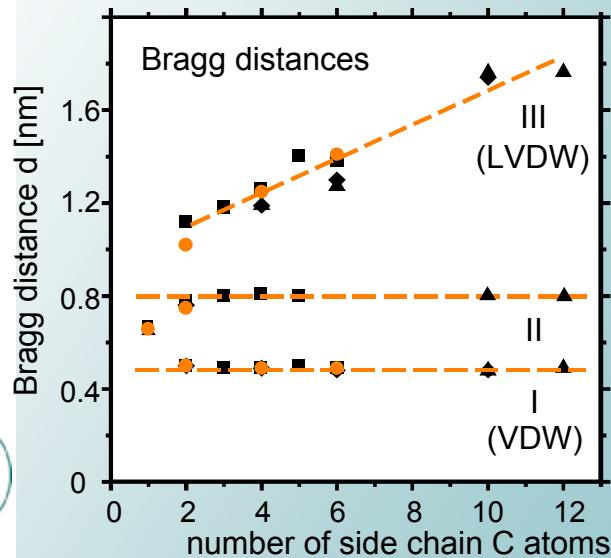
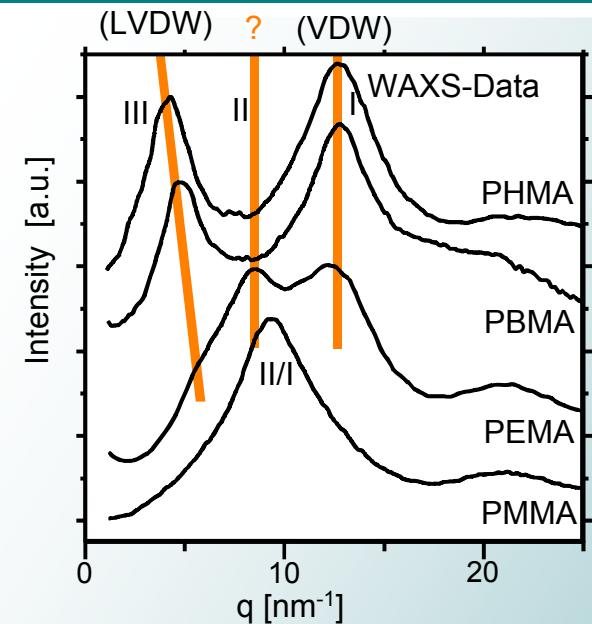


$$M_w = 0,46 \text{ kg mol}^{-1}$$

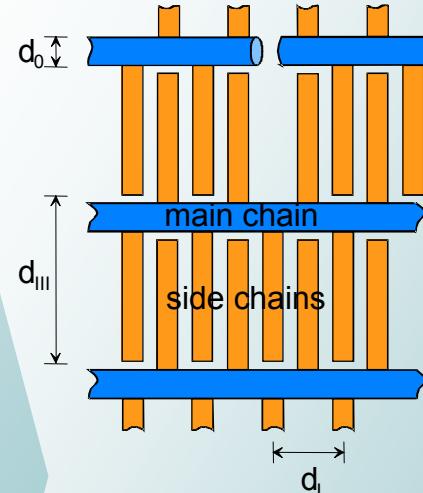
$$M_w/M_n = 1,14^*$$



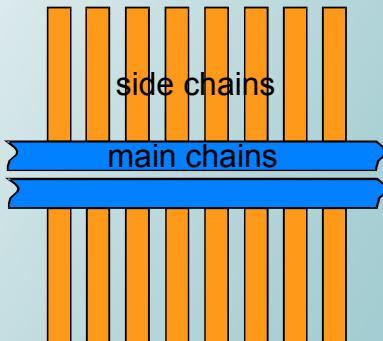
# Organisation in Poly(Methacrylates): WAXS



syndiotactic poly(methacrylate)

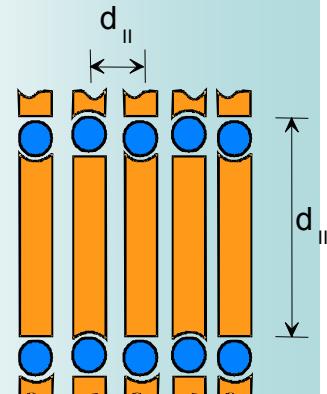


isotactic PEMA



extrapolated lokal structur:

**"Nano Layers"**





# **Investigation of Structure and Dynamics in Polymeric Systems via Solid State NMR**

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**Introduction** • Interaction in solid state NMR

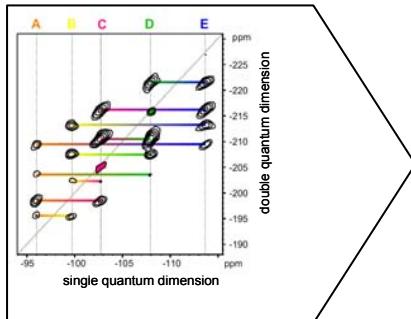
**Solid State NMR** • MAS, recoupling, double-quantum NMR

**Polymer Dynamics** • polyelectrolyte layers, polybutadiene, PEMA

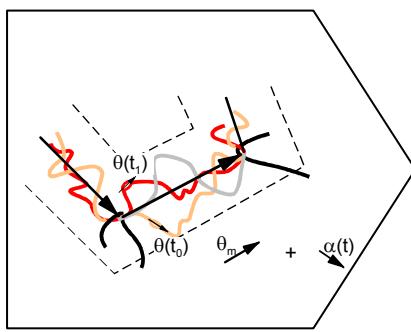
**Conclusions** • solid state NMR investigations



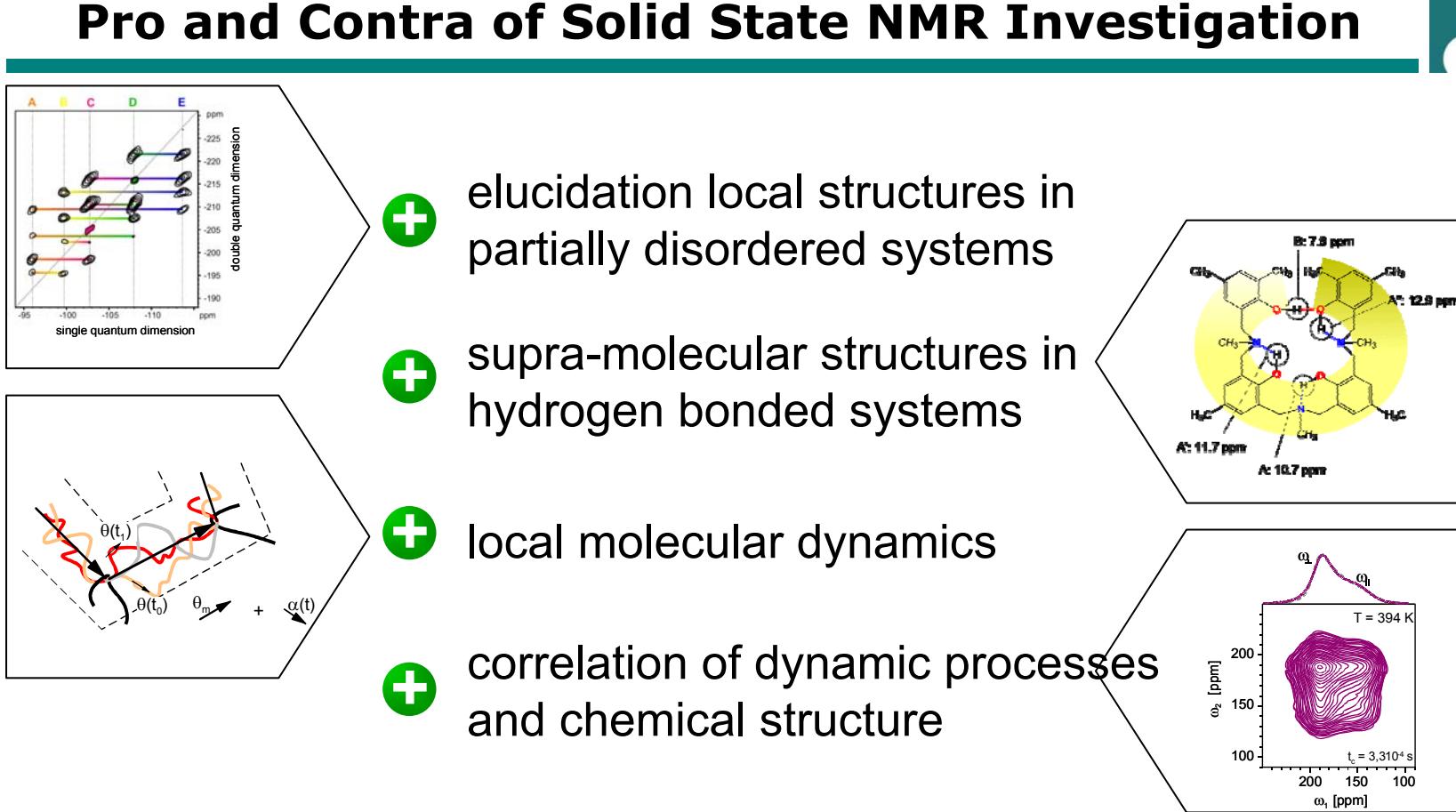
# Pro and Contra of Solid State NMR Investigation



+ elucidation local structures in partially disordered systems

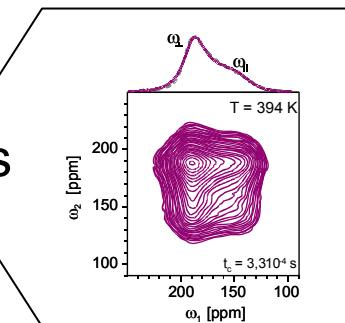


+ supra-molecular structures in hydrogen bonded systems



+ local molecular dynamics

+ correlation of dynamic processes and chemical structure



- needs expertise

- expensive

- ...



# Acknowledgements

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Prof. Dr. H. W. Spiess (\$, €, ...)

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Prof. B. Chmelka, Dr. N. Hedin (layered silicates)

Prof. K. Ishida, Dr. D. Sebastiani (polybenzoxazines)

Prof. L. Reven, Dr. M. McCormick (polyelectrolytes)

Prof. A. Heuer (polymer theory, ...)

Dr. T. Dollase, Dr. M. Neidhöfer (polybutadiene)

Dr. M. Wind, Dr. W. Steffen, Prof. Do Yoon (PEMA)

